Experimental research with exotic nuclei

Vratislav Chudoba

FLNR, JINR

Periodic table

дон	Ħ									групп	ыэ	лемен	ITOB							1					
uep	5d	a	Ιđ	a	II	б	a	Ш б	a	IV	б	a	V	б	a	VI	б	a	VII	б	a	VIII	б		
1	I	Водород Н 1,00794 Hydrogen	1 1s'																		Гелий Не 4,0026 Helium	$\frac{2}{1s^2}$			
2	п	Литай Li 6,941 Lithium	3 2s'	Берили Ве 9,0121 ВегуШ	ий 4 2s ²		Бор В 10,811 Вогоп	5 2p'	Yrnepoj C 12,011 Carbon	6 2p ²		Asor N 14,00674 Nitrogen	7 2p ³		Кислород О 15,9994 Охудеп	8 2p ⁴		Фтор F 18,998403 Fluorine	9 2p ⁵		Неон Ne 20,1797 Neon	10 2p ⁶		Водород 1 Н 1s ¹	
3	ш	Harpui Na 22,98976 Sodium	11 3s ¹	Marsue Mag 24,305 Magne	12 3s ²		All All 26,981539 Aluminum	^a 13 3p'	Кремни Si 28,0855 Silicon	[#] 14 3p ²		Фосфор Р 30,973762 Phosphorus	15 3p ³		Cepa S 32,066 Sulfur	16 3p ⁴		Xnop Cl 35,4527 Chlorine	17 3p ⁵		Аргон Аг 39,948 Argon	18 3p ⁶		Hydrogen символ / symbolатом	алектронная конфигурация / electronic configuration ная масса / atomic mass
4	IV	Калий K 39,0983 Potassium	19 4s'	Kaman Ca 40,078 Calciur	^{di} 20 4s ²			21 3d ¹ 4s ² Sc 44,955910 Scandium		22 3d ² 4s ²	Титан Ti 47,88 itanium		23 3d ³ 4s ²	anagmä V 60,9415 nadium		24 3d ⁵ 4s ¹	Xpom Cr 51,9961 romium		25 ^M 3d ⁵ 4s ²]	apraneu Mn 4,93805 nganese		26 3d ⁶ 4s ²	железо Fe 55,847 Iron	27 3d ⁷ 4s ² C0 58,93320 Cobalt	28 3d ⁸ 4s ² Ni 58,6934 Nicket
-	v		29 3d ¹⁰ 4s ¹ Cu 63,54 Copp	њ б ат	30 3d ¹⁰ 4s	² Zn 65,39 Zinc	Fannski Ga 69,723 Gallium	31 4p ¹	Герман Ge 72,61 German	^{dit} 32 4p ²		Mannask AS 74,92159 Arsenic	33 4p ³		Селен Se 78,96 Selenium	34 9.7 4p ⁴	5238	Бром Br 79,904 Bromine	35 4p ⁵		Криптон Kr 83,80 Krypton	36 4p ⁶		S-ЭЛЕМЕНТЫ/	ELEMENTS
5	VI	Pyбяднё Rb 85,4678 Rubidium	37 5s ¹	Crpost Sr 87,62 Strontin	анй 38 5s ²			39 4d ¹ 5s ² Х 88,90585 Учитиал		40 ⁴ 4d ² 5s ² Zin	рконий Zr 91,224 rconium		41 ¹ 4d ⁺ 5s ¹] 92 N	Inofinit Nb ,90638 iobium		42 ^{Mo} Id ⁵ 5s ¹ I Molyl	анбден Mo 95,94 bdenum		43 ^{Te} 4d ⁵ 5s ² Tec	траниций Tc [98] hnetium		44 4d ⁷ 5s ¹	Рутений Ru 101,07 uthenium	45 4d ⁸ 5s ¹ Rh 102,90550 Rhodium	46 ^{Палладий} 4d ¹⁰ Рd 106,42 Palladium
5	VII		47 4d ¹⁰ 5s ¹ Ag 107,868 Silve	0 5 2 1	48 4d ¹⁰ 5s	Kagsouli Cd 112,411 Cadmiuan	In In 114,818 Indium	49 5p'	Олово Sn 118,710 Tin	50 5p ²		Сурьма Sb 121,757 Antimony	51 5p ³	-	Tennyp Te 127,60 Tellurium	52 5p ⁴		Иод I 126,90447 Iodine	53 5p ⁵		Ксенон Хе 131,29 Хепоп	54 5p ⁶		 d-элементы f-элементы 	
6	vш	Цезий CS 132,90543 Cesium	55 6s'	Барий Ва 137,327 Вагіит	56 6s ²			57 5d ¹ 6s ² La 138,9055 Lanthanum		72 5d ² 6s ²	Гафний Hff 178,49 Iafnium		73 5d ³ 6s ²	Taerran Ta 0,9479 ntalum		74 ^{Bo.} 5d ⁴ 6s ²	њфрам W 183,84 ungsten		75 5d ⁵ 6s ²	Рений Re 186,207 thenium		76 5d ⁶ 6s ²	Ocmuñ OS 190,23 Osmium	77 Ирилий 5d ⁷ 6s ² Ir 192,22 Iridium	78 5d ⁹ 6s ¹ Рt 195,08 Platinum
U	IX		79 5d ¹⁹ 6s ¹ AU 196,9665 Gol	0 L 4 d	80 5d ¹⁰ 6s	Pryrs Hg 200,59 Mercury	Tautus Tl 204,3833 Thallium	81 6p ¹	Свинец Pb 207,2 Lead	82 6p ²		Висмут Ві 208,98037 Bismuth	83 6p ³		Polonium	84 6p ⁴	*	Actar At [210] Astatine	85 6p ⁵		Радон Rn [222] Radon	86 6p ⁶			
7	x	Франций Fr [223] Francium	87 4,073 7s ¹	Pagnii Ra 226,02 Radium	88 7s ²			89 6d ¹ 7s ² Асс [227] Actinium		104 ^{Pesep} Ruther	popratii Rf [261] fordium		105]] Di	ly6ancii Db [262] ibaium		106 ^{Cu} Seab	боргий Sg [266] orgium		107	Борий Bh [267] Зоhrium		108	Хассий HS [269] Hassium	109 ^{Мейтнерий} Мt [268] Meitnerium	110 Дармштадтий Ds [269] Darmstadtium
<i>'</i>	XI	1	11 Permenti Rg I272 Roentgenius		112	Konepsiekunik Cn [285] Copernicium		113	Флерон Fl Flerovi	^{вий} 1	14		115		Ливермој Lv Livermori	^{инй} 1	16		117		118				
Лан	ганои	цы I	anthanides																						
Цери Се 140,119 Сегішт	t 4f'5d'	II F 14 Pr	разеодим Г 4f ⁴ 0,90765 ascodymiam	Неоди Nd 144,24 Neoxlym	м 4f ^a ium	Pm [145] Prometh	етий l 4f ⁴ hium	Самарий Sm 4f ⁶ 150,36 Samarium	1	Европий Ец 4f [°] 151,965 Евгоріцт		Гадолиний Gd 4f ³ 5 157,25 Gadolinium	i d'	Тербиі Tb 158,9253 Terbium	i 4f" 14	Диспр Dy 162,50 Dyspros	озий 4f ¹⁰ ium	Fo Ho Ho	льмий HO 4f ¹¹ 4,93032 olmium		Эрбий Er 4f ^d 167,26 Erbium	2	Тулий Tm 4 168,93421 Thulium	Иттербий 4f ¹³	Лютеций Luu 4f ¹⁴ 5d ¹ 174.967 Lutetium
AKT	иноид	цы	Actinides																						
Topui Th 232,038 Thoria	7s ² 6d ²	II P 23 Pr	ротактиний а 51°6d' 1,03588 Mactinium	Уран U 5f 238,028 Uranium	°6d'	Henry Np [237] Neptani	лний 5f ⁴ 6d ¹ ium	Плутоний Pu 51 ⁶ [244] Plutonium		Америций Am 5f [°] (243) Americium		Кюрий Cm 5f [°] [247] Curium	6d'	Беркли Bk [247] Berkeliur	ій 5 f ^{°°} m	Калиф Cf : [251] Californ	юрний 5f ¹⁰ ium	Эi Ei Ei	йнштейний SS 5f ¹¹ s2j nsteinium		Фермий Fm 5 [257] Fermium	if ¹²	Mенделе Md : [258] Mendelevin	свий Нобелий 5f ¹³ No 5f ¹⁴ um Nobelium	Лоуренсий Lr 5f ¹⁴ 6d ¹ [262] Lawrencium

Nuclear landscape



Why exotic?

- far from stable nuclei
- complicatedly available



too heavy or strange in other way

What's going on?

Modest vs. big, bigger, the biggest





Modest vs. big, bigger, the biggest

SPI Facility for **Antiproton and Ion Research:** FRIB •10 member states scientists from more than 50 countries 20 years of construction **RIBF** 1,6 billions euro Para PL Pi

Experimentier- und Speicherringe

SHARAO

Radioactive ion beams

acceleration of a primary beam (I~10¹² pps)

Radioactive ion beams



Radioactive ion beams

acceleration of a primary beam (I~10¹² pps)

ISOL technique

- reactions in a thick production target: (fast production – slow release)
- reaction products to be extracted, ionized and reaccelerated
- secondary beam: (I<10⁸ pps)



reactions on a physical target



secondary beam: fragment-separator (I<10⁶ pps)

reactions on a physical target

Examples of RIB facilities

- In flight
 - GANIL (France), RIKEN (Japan), GSI (Germany), MSU (USA)
- ISOL
 - REX-ISOLDE (CERN), SPIRAL (France), TRIUMF (Canada)
- Other
 - Sao Paolo (Brazil), Orsay (France),
 Catania (Italy), Oak Ridge (USA),
 Jyvaskyla (Finland), Dubna

FLNR itself



Nuclear physics in FLNR

Heavy and superheavy elements

- synthesis of superheavy nuclei
- nuclear spectroscopy
- mass spectrometry
- Light exotic nuclei
 - properties and structure of light exotic nuclei
 - reactions with exotic nuclei

FLNR accelerator complex Current state of FLNR



FLNR accelerator complex

Full-scale realization of the DRIBs-III

Dubna Radioactive Beams



Superheavy elements factory



SHE: Island of stability



SHE: to the Island of Stability

synthesis of superheavies

$$Z_1 + Z_2 = Z$$

 $N_1 + N_2 = N + (2 - 4)n$

"cold" fusion: Pb + heavy ion "hot" fusion: light beam + heavy target





- low-energy physics
- compound nucleus
- combination of light and heavy nuclei gives higher cross sections

SHE: synthesis of superheavies



Periodic table (~150 years ago)

Onbunes are menter successed B; conobastallo un anaunicut che a goden Ba, D. Menceneta. Ji=so Ex=90 ?= 180. V= 51 No=94 Ja=189. G=52 Mo-16 W= 186. Mass Rh=109,4 Pt=197,4. Fe=56 Ro=109,4 De=193. $\begin{array}{c} \mathcal{F}_{e=55} & \mathcal{R}_{o}^{i=109} & \mathcal{F}_{e=193} \\ \mathcal{K}_{e} = 53 & \mathcal{R}_{e=1056} & \mathcal{C}_{e+19} \\ \mathcal{K}_{e} = 53 & \mathcal{R}_{e=57} & \mathcal{R}_{e=1056} & \mathcal{C}_{e+19} \\ \mathcal{K}_{e=59} & \mathcal{K}_{e=29} & \mathcal{C}_{e=63,9} & \mathcal{L}_{g=240} & \mathcal{K}_{g=240} \\ \mathcal{K}_{e=59} & \mathcal{K}_{e=29} & \mathcal{K}_{e=657} & \mathcal{C}_{e=112} & \mathcal{K}_{g=240} \\ \mathcal{K}_{e=59} & \mathcal{K}_{e=279} & \mathcal{K}_{e=68} & \mathcal{K}_{e=106} & \mathcal{K}_{e=197} \\ \mathcal{K}_{e=12} & \mathcal{K}_{e=23} & \mathcal{K}_{e=279} & \mathcal{K}_{e=118} \\ \mathcal{K}_{e=10} & \mathcal{K}_{e=23} & \mathcal{K}_{e=29} & \mathcal{K}_{e=128} \\ \mathcal{K}_{e=12} & \mathcal{K}_{e=23} & \mathcal{K}_{e=39} & \mathcal{K}_{e=29} & \mathcal{K}_{e=128} \\ \mathcal{K}_{e=10} & \mathcal{K}_{e=23} & \mathcal{K}_{e=39} & \mathcal{K}_{e=372} & \mathcal{K}_{e=279} \\ \mathcal{K}_{e=10} & \mathcal{K}_{e=237} & \mathcal{K}_{e=39} & \mathcal{K}_{e=277} & \mathcal{K}_{e=277} \\ \mathcal{K}_{e=10} & \mathcal{K}_{e=278} & \mathcal{K}_{e=278} \\ \mathcal{K}_{e=278} & \mathcal{K}_{e=278} & \mathcal{K}_{e=278} & \mathcal{K}_{e=278} \\ \mathcal{K}_{e=278} &$ G= 55? da= 94 ? yt= 60? Si=95 ? Sn = 750? Sh=118? Essai d'une des des de ments d'après leurs poids atomiques et fonctions chimiques fran & ellendelles protes de des ellendelles protes de des ellendors à l'Alendelles protes de des ellendors de l'annes de la des protes de des ellendors de la des de la des protes de des ellendors de la des de la des protes de de des de des de la des de la de de la des de la de la de de la de de la de la de la de de la d 18 <u>II</u> 69. I yracy bedruch reages honorur lou verfuo de писано, по гаокача mohichere usino. audurch noting to to anonten to bab repuys by



Periodic table today

			III4 Fleroviu	Dubna		Tab		ELAGUVAGE									
1	110	priv	Эди	ACCU	heads -		N. C.	a de se	- A	18							
Водород 1 ₁₅ Н ^{13,29841} 0,0890 1,00794 -229,37	2					4	Ц.И.	Me	нде	nee	ва	13	14	15	16	17	Телий 2 1с Не 24.58724 8.1785 4.0026 - 2722 Нейом 268.8
Jarnii 3 3/2 Li 5.29175 5.39175 6,941 150,941 150,941 Lithium 1542 1542	Бериллий 4 2, Ве 9,3226 1841 9,01218 128 Beryllium 247	D.I.	Men	delee	ev's F	Perio	dic 1	Table	of E	leme	ents	Бор 5 ₂₇ В 32% 10,811 3075 Вогол 4000	Углерод 6 25 ⁷ С 11,970 12,011 Carbon ^{(2,00} ,3657	Asor 7 20' N 14,52414 14,0067 -23,000 Nitrogen 195,79	Кислород 8 _{2р} О ^{13/18} 15,9994 -218,75 Охудеп ^{13/2}	Фтор 9 27 F 15,42283 18,9984 - 215,62 Fluorine 188.12	неон 10 ₂₆ ³ Neon 20,1797 - 248,56 Neon - 215,45
Harpstii 11 36 Na 5.13907 22,989768 97.73 Sodium 882	Магний 12 Mg 7,6162 24,3050 653 Magnesium 1000	3	4	5	6	7	8	9	10	11	12	Алюминий 13 _г Al 598535 26,981539 6932 Aluminum 2515	Кремний 14 ₂₅ Si 5500 28,0855 5ilicon 3255 14 ± 3265	Фосфор 15 ₃₀ ² Р ^{III.48869} 1820 30,97376 44.15 Phosphorus 777	Cepa 16 36 5 32,066 015,20 5 32,066 015,20 5 32,066 015,20 5 32,066 015,20	хлор 17 уў Cl ^{12,963742} 35,4527 - 401,5 Chlorine - 34.04	Аргон 18 _{3р} : Ar (17857 39,948 -499,35 Argon -485,85
Калий 19 ₄₅ К ^{1,1/066} ⁸⁶³ 39,0983 ^{65,38} Potassium ⁷³⁹	Кальций 20 ₄₅ Са 6.11316 540 40,078 812 Calcium 484	Скандий 21 34 [°] bc 500 (5591) 500 (54) 500 (54)	титан 22 76 с. 30 47,88 1 тапіцт 2287	Ванадий 23 _{3d-6} хр. V ^{6,7469} 50,9415 1910 Vanadium 3007 Chi	1078 24 1018 6,5664 7200 1,9961 1916 1000 100	Марганец 25 Mn ^{7,43400} 7200 54,93805 7346 Manganese 2061	железо 26 36% Fe 7568 55,847 1538 Гоп 2561	Кобалыг 27 Луча СО ^{5,840} 58,93320 1465 Соbalt 2827	никель 28 _{30%} Ni 2002 58,6934 1455 Nickel 2913	медь 29 Си ^{5/2/33} 63,546 084/0 Соррег 2562	цинк 30 30'45' Zn ^{3,39455} 55,39 4 9,33 Zinc 907	Галлий 31 ₄₇ Ga 55930 594 69,723 29,76 Gallium 3334	Германий 32 _{4,} Ge ^{7,500} 72,61 ^{938,25} Germanium ^{938,25}	Мышьяк 33 45 As 25.111 74,92159 Ст. 6.4 Arsenic	Селен 34 _{-р} Se ^{9,78278} ⁴⁵¹⁰ 78,96 221 Selenium 985	Бром 35 45 Вг ^{11,81581} 79,904 7.2 Bromine ^{38,73}	Криштон 36 _{2р} Кг ^{3,596} 1 83,80 187,36 Ксурton -159,22
рубидий 37 ₅₆ Rb - 179137 85,4678 - 9,3 Rubidium - 588	Стронций 38 55 Sr 25448. 87,62 77 Strontium 1383	Иттрий 39 4ď36 Y 4ď36 169 88,90585 15% Yttrium 3327	цирконий 40 ₆₆₅₅ Zr 6 30 91,224 30 2 irkonium 4 35	ниобий 41 ₄₅₅ Мо Nb 6/9888 92,90638 муу Niobium 2314 Мо	олибден 42 ₁₈₅₈ 10 уноча 1000 5,94 леза olybdenum - 629	Технеций 43 ₄₅₅₆ Тс ⁷²⁸ 11489 [98] ²¹⁶⁷ Technetium 4265	Рутений 44 ибл. 2,4080 101,07 234 Ruthenium 4150	Родий 45 Rh 12400 102,90550 1944 Rhodium 5655	Палладий 46,3° Pd 82555 106,42 15555 Palladium 2555	Серебро 47, 48 ⁴⁰ 28 ⁴ Ад 20,0634 10500 107,8682 Silver 2162	Кадиний 48 _{16,55} ; Cd ^{8,29,66} 112,411 2529 Cadmium 252	индий 49 ₅₅ In 5.78656 114,818 1.55.6 Indium 3072	Олово 50 ₅₅ Sn ⁷⁵⁴³⁸¹ ^{118,710} 231.93 Tin 2002	сурьма 51 ₅₀ ; Sb 8.64 121,757 610.61 Antimony 1587	Теллур 52 _{5р} Те 80096 6556 127,60 449,51 Tellurium 998	Иод 53 55 I 10/ \$128 4990 126,90447 113,7 Lodine 184,4	ксенон 54 ₅₅ ° Xe ^{12,12987} 131,29 Xenon ^{11,75}
цехий 55 6 CS ^{1,89360} 132,90543 23,44 Cesium 671	Bapmii 56 Ba 521 7 137,327 72 Barium 159	Лантан 57 La 5,5750 5149 138,9055 520 Lanthanum 3454	Гафний 72 ₅₆₅ Нf 68397 12310 178,49 12310 Наfnium 4893	Тантал 73 ₅₄₆ ; Во. Та ^{2,68} 180,9479 1617 Тапtаlum 5455	альфрам 74 _{3.765} Л 2.98 19309 13,84 4122 ingsten 5355	Рений 75 Re 21530 186,207 3186 Rhenium 5536	осмий 76 56%; 0s 23450 190,23 2012	иридий 77 Sries 192,22 Iridium 468	платина 78 _{50%} Pt ^{9,0} 195,08 15%4 Platinum 2525	Залото 79 _{51[°]бе¹} Au ^{9,22567} 19910 196,96654 064, 8 Gold 2559	Ртуть 80 _{50.165} Нд 10.44750 135463 200,59 -38.81 Метситу 359,73	таллий 81 _{sp} Tl 5.10525 204,3833 391 Thallium 471	Свинец 82 ₆₃ Рb ^{7/1665} 10350 207,2 327.45 Lead 1749	висмут 83 _{ір} . Ві ^{7,289} 208,98037 208,98037 271,1 Bismuth ¹⁵⁶⁴	Нолоний 84 _{бр} Ро ^{8,41671} [209] 251 Polonium 952	Actar 85 cc At 90 [210] 902 Astatine 337	Радон 86 _{бр} . Rn ^{0,71850} [222] -71 Radon -2013
Франций 87 к Fr 4,013 [223] 27 Francium 607	Радий 88 ул Ra 5,2389, 9401 226,025 то Radium 1.44	Актиний 89 9/33 Ас (227) 1030 Астівіцт 303	Резерфордий 104 Rf of [261] Rutherfordium	Дубиай 105 См Db [262] Dubnium 60 S [262] Dubnium	нборгий 106 66] aborgium	Борий 107 Bh ⁶² [262] Bohrium	Хассий 108 Hs (*' [269] Hassium	Мейтнерий 109 Mt rd [268] Meitnerium	Дармитадтий 110 DS (d [269] Darmstadtium	Рентгений 111 Rg [272] Roentgenium	Консринкий 112 Сп [285] Сореглісіцт	нихоний 113 Nh Nihonium	Флеровий 114 Fl Flerovium	масковий 115 Mc Moscovium	ливерморий 116 LV Livermortum	Теннессик 117 TS Tennessine	Oranecon 118 Og Oganesson

Лантаноиды Lanthanoides

Церий	58 ₄₁₅₄	Празеодим 59	Неодим	60 ₁₁	Прометий С	51 🔐	Самарий	62 ₋₀	Европий	63 ₁₁	Гадолиний (64 ₁₁₂₄	Тербий	65 ₄₁	Диспрозий	66 _a ,	Гольмий	67 🦼	Эрбий	68 ₄₁	Тулий	69 _e	Иттербий	70 ₋₁₁	Лютеций	71	
Се	5.5387 6778	Pr 5.47	Nd	5,525 7009	Pm	2.35 726-	Sm	5,6437 7520	Eu	3.6704 5244	Gd	6.130 3901	Tb	5,8639 \$230	Dy	3.9389 8551	Но	6.02 S	Er	6,1038 9066	Tm	6.18431 9321	Yb	6.254-6 (965	Lu	5,42595 084]
140,115 Cerium	7519 3424	140,90765 99 Praseodymium 3510	144,24 Neodymiur	1016 n 5060	[145] Promethium	1042 3003	150,36 Samarium	1623 1750	151,965 Europium	822 1596	157,25 Gadolinium	1314 3364	158,92534 Terbium	1859 3221	162,50 Dysprosium	1411 3561	164,93032 Holmium	1472 2694	167,26 Erbium	1509 2683	168,93421 Thulium	546 1946	173,04 Ytterbium	824 19	174,967 Latetium	1662 3393	1

Актиноиды Actinoides

Торий	90 _{5%}	Протактиний93	Jred Y	^{уран} 92 ₃	нептуни	nii 93 ₅₍₆₁	Плутоний	94 _{sc}	Америций	95 _я	Кюрий	96 _{scod}	Берклий	97 _{s:}	Калифорний 9	98,,,,	эйнштейний 99	Ферми	100	менделевий101	s"	Нобелий 102	58	лоуренскії 103
Th	6/08 11700	Pa	3.89 15370	U al	Np	6.2657 20250	Pu	6.95 19840	Am	5,952 13670	Cm	6.02	Bk	6.23 11780	Cf	6.30	Es se	Fm		Md °	38	No «		Lr :
232,0381 Thorium	1750 4790	231,03588 Protactinium	1572 2 - U	38,0289 Iranium	135 [237] 191 Neptunii	644 ₩197	[244] Plutonium	640 3328	[243] Americium	1126 2607	[247] Curium	1345	[247] Berkelium	1050	[251] Californium	960	[252] 50 Einsteinium	[257] Fermiur	n	[258] ^s Mendelevium	827	[259] s Nobelium	\$27	262] Lawrencium 1627



H - custoar, / symbol 1.00794 - arownen wacca / atomic mass 131 - anerpowen workpuryacus / electron configuration 13,59844 - 1-in normulaa workacuus, 36 / 1st lonization potential, eV 0.0899 - normore, sk / wk / density, kej / mk - 253,34 - rewneparypa musewen, °C / boling temperature, °C - 252,87 - rewneparypa musewen, °C / boling temperature, °C

ACCULINNA



ACCULINNA-2



- energy range 6 60 MeV/A
- beam intensities higher in 2 orders

•
$$Z_{\rm RIB} \sim 1 - 36$$







New megascience project



New megascience project

Concept of ERICA complex

(Electron - Radioactive Ion Collider Assembly)



Why exotic?

- far from stable nuclei
- complicatedly available





Drip-line



moving away from the nuclear stability \rightarrow transition from discrete spectrum to continuum \rightarrow immediate emission of nucleon

Nuclear halo



Nuclear halo



- tunneling to the forbidden regions
- extended size of nucleus

B. Jonson P.G. Hansen. The Neutron Halo of Extremely Neutron-Rich Nuclei. Europhys. Lett., 4(4):409–414, **1987**

Nuclear halo



First observation of ⁶He

T. Bjerge. Radio-Helium. NATURE, 137, 865, 138:400–400, **1936**!!!

Observation of large ⁶He radius

I. Tanihata et al., Physics Letters B, 160(6):380– 384, 1985.

Borromean nuclei



Soft dipole mode (SDM) of Giant dipole resonance (GDR)



GDR

- protons vs. neutrons
- $E_{GDR} \sim 14 24 \text{ MeV}$
- induced by EM excitation
Soft dipole mode (SDM) of Giant dipole resonance (GDR)



GDR

- protons vs. neutrons
- $E_{GDR} \sim 14 24 \text{ MeV}$
- induced by EM excitation

SDM

halo vs. core



- E_{SDM} lower than E_{GDR}
- induced by EM excitation and chargeexchange reaction

Soft dipole mode (SDM) of Giant dipole resonance (GDR)

GDR

- protons vs. neutrons
- $E_{GDR} \sim 14 24 \text{ MeV}$
- induced by EM excitation

p

SDM

- halo vs. core
- E_{SDM} lower than E_{GDR}

2 n

 induced by EM excitation and chargeexchange reaction

Radioactivity and decays



α decay E. Rutherford, 1899

β⁻ decay H. Becquerel, 1886

β⁺ decay F. and I. Joliot-Curie, 1932

p rad. S. Hoffman, 1982 **2p rad.** M. Pfützner, B. Blank, 2002 **cluster emission** H. Rose, 1984

(multi)-n radioactivity still waiting

Proton radioactivity



p-radioactivity

natural generalization of α-radioactivity



Proton radioactivity



natural generalization of α-radiactivity

2p-radioactivity no analogue in classical mechanics





Proton radioactivity



p-radioactivity natural generalization of α-radiactivity

2p-radioactivity

no analogue in classical mechanics



⁴⁵Fe, ¹⁹Mg, ⁵⁴Zn, ⁴⁸Ni, ⁶⁷Kr, ^{94m}Ag

Neutron radioactivity



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General tasks

existence of the isotope
mass of the isotope
decays
eventually J^π identification









Mass measurement

- mass spectrometers
- calculations from decay kinematics





Decays



Decays



3-body decays

2-body vs. 3-body decay

- 2 parameters for 2-body decay (E,Γ)
- 5 additional parameters at given energy for 3-body decay



Correlations

• full description of the internal correlations by parameters ε and θ_{k}

$$\varepsilon = \frac{E_x}{E_x + E_y}$$

-

$$\cos\theta_k = \frac{\mathbf{k}_x \cdot \mathbf{k}_y}{k_x k_y}$$

external correlations:
3-body system
orientation



Correlations

• full description of the internal correlations by parameters ε and θ_{k}



External correlations



Available experimental methods

- ion-implantation method
- decay-in-flight by tracking technique
 - -information on life-time accessible
 - -identification of 2p-decay channels by correlations

Optical Time Projection Chamber



Optical Time Projection Chamber



Decay-in-flight by tracking



Distance from target to decay vertex

Life-time measurement by tracking

- characteristic shape of vertices distribution
- suitable for lifetimes $10^{-7} 10^{-12}$ s







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- transition $k_{p-HI} \rightarrow \theta_{p-HI}$
- without measurement of proton energies

I. Mukha et al. Phys. Rev. C 82 (2010) 054315 ¹⁶Ne



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ACCULINNA experiment



ACCULINNA experiment



⁶Be: Invariant mass spectrum



⁶Be: Internal correlations



Theoretical model

PWIA in combination with 3-body model

- test on 0⁺ ground state (no free parameters)
- overlapping states 0⁺ and 2⁺
 - □ 2⁺ alignment
 - \Box Interference of 0⁺ and 2⁺

low sensitivity of internal correlations to model parameters

⁶Be: Quasibinary kinematics


⁶Be: External correlations



- dramatic changes of the θ_{α} distributions depending on the model parameters
- level of alignment and interference angle is changing with E_T and θ_{Be}



is changing with E_{τ} and $\theta_{\rm Be}$

analysis was done for the first time

¹⁰He: Missing mass spectrum



$${}^{3}\mathrm{H} + {}^{8}\mathrm{He} \rightarrow {}^{10}\mathrm{He} + \mathrm{p}$$

 ${}^{10}\mathrm{He} \rightarrow {}^{8}\mathrm{He} + \mathrm{n} + \mathrm{n}$

- missing mass spectrum from protons measured in coincidence with ⁸He
- 479 events found

S.I. Sidorchuk et al. Phys. Rev. Lett. 108 202502 (2012)

- population of 0+ ground state with maximum at E_{τ} ~2.1 MeV
- structureless spectrum for $E_T > 4$ MeV

¹⁰He: Correlations



• J^{π} is fully determined by L of ⁸He

$$\varepsilon = E_{\rm nn} / E_T^{\rm (He)} < 0.5$$

¹⁰He: Correlations



- J^{π} of the ground state confirmed by the experimental data analysis
- J^{π} of the **1**⁻ states determined from experimental data for the first time

Conclusion and outlook

- extensive field for pioneering research
- 3000 isotopes known, 3000 to be discovered
- new RIB factories constructed worldaround
- new facility in Dubna, new plans for the future (ERICA)
- new research methods

Thank for attention



Spin-parity identification

Standard methods:

- Elastic resonance scattering
- Direct reactions

Elastic resonance scattering

- fit to R-matrix
- \bullet unambiguous J^{π} identification
- constrains:
 - width of resonance
 - existing beams

Direct reactions

- angular distribution of products
 - using (e.g. DWBA) ΔL may be determined
- "zero geometry" approach

Zero geometry approach



Zero geometry approach





Zero geometry approach

- constrains:
 - spinless reaction participants
 - high reaction cross section needed
- it can be easily generalized
 → correlation patterns















- neutron radioactivity studies
- $E_n \sim 200 800$ MeV in LAB
- low transverse momenta
 0.1 100 keV
- precise information on angular correlations of decay neutrons with a charged fragment
- angular resolution ~0.1 0.2 mrad





28 m from the target in FMF2

at least 36 modules 30 x 30 x 100 cm³

bundles

- 3 x 3 mm scintillation fibers BCF12
- 48 x 48 x 1000 mm
- 2 MAPMT from both sides





- **longitudinal coordinate** of the n interaction along the fiber
- determination the very first hit
- avoid neutron cross-talk



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Appendix: Nuclear halo

Stable nuclei

 $< r_n^2 >^{1/2} - < r_n^2 >^{1/2} \approx 0.1 fm$

Exotic nuclei

$$< r_n^2 >^{1/2} - < r_p^2 >^{1/2} \gtrsim 1.5 fm$$

neutron halo

one neutron: ¹¹Be, ¹⁹C two neutron: ⁶He, ¹¹Li, ¹⁷B, ¹⁹B, ²²C neutron skin: ⁸He and ¹⁴Be

proton halo

g.s. of ⁸B, ¹³N, ¹⁷Ne, ²⁶P, ²⁷S the first e.s. of ¹⁷F

Appendix: Dipole modes



resonance vs.

mode

- property of particular nucleus
- its population does not depend on reaction mechanism

- characteristic for specific reaction
- its population is given by reaction mechanism

Appendix: microStrip detectors



Hardware

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Appendix: Theoretical model

- PWIA in combination with 3-body problem
- task reduced to solving of Schroedinger equation with source

$$(\hat{H}_3 - E_T)\Psi_{^{6}\text{Be}}^{JM(+)} = \hat{\mathcal{O}}_{\mu'\mu}\Psi_{^{6}\text{Li}}^{J^{(\text{in})}M^{(\text{in})}}$$

 transition operator contains information about population of 6Be from 6Li

$$\hat{\mathcal{D}}_{\mu'\mu} = \sum_{i=1,2} \sum_{lm} f_l(q, r_i) Y_{lm}(\hat{r}_i) Y_{lm}^*(\hat{q}) \ \tau_-^{(i)} \ \sum_{\nu} (-1)^{\nu} \sigma_{\nu}^{(i)} C_{\frac{1}{2}\mu'\nu}^{\frac{1}{2}\mu'\nu}$$

• Transition operator takes a "simple" analytical form thanks to the choice of the N-N potential used in PWIA $\hat{V}_{ir}(r_{ir}) = (\boldsymbol{\sigma}_i \cdot \boldsymbol{\sigma}_r)(\boldsymbol{\tau}_i \cdot \boldsymbol{\tau}_r) V_0 \exp\left[-\left((\mathbf{r} + \mathbf{r}_i)^2 / r_0^2\right)\right]$

Three-body

, r_{1r}

r_{2r}

r₁₂

r_{α1}

r α2

Ω
NUSTAR

NUclear STructure,

Astrophysics and Reactions

 HISPEC/DESPEC (High-Resolution Spectroscopy/Decay Spectroscopy)



- R3B (Reactions with Relativistic Radioactive Beams)
- MATS (Precision Measurements of very short-lived nuclei with Advanced Trapping System)
- LaSpec (Laser Spectroscopy)
- ILIMA (Isomeric Beams, Lifetimes and Masses)
- ELISe (Electron-Ion Scattering in a Storage Ring)
- EXL (Exotic nuclei studied in light-ion induced reactions at the NESR storage ring)

Super-FRS Experiments

SHE (Super-Heavy Element Research)

SuperFRS Experiments

- Mass and charge resolution
 - Search for new isotopes and ground-state properties
 - Atomic collisions
- Unique experiments at Super-FRS as high-energy high-resolution spectrometer
 - Spectroscopy of meson-nucleus bound system
 - Exotic hypernuclei and their properties
 - Importance of tensor forces in nuclear structure
 - Delta resonances probing nuclear structure
- Experiments taking advantages of multi-stages and high-resolution of the Super-FRS
 - Nuclear radii and momentum distributions
 - EXPERT (EXotic Particle Emission and Radioactivity by Tracking)
 - Low-q experiments with an active target
 - Nuclear reaction studies and synthesis of isotopes with low-energy RIBs