HELMHOLTZ INTERNATIONAL SUMMER SCHOOL

Dubna International Advanced School of Theoretical Physics / DIAS-TH

DENSE MATTER 2015

Bogoliubov Laboratory of Theoretical Physics, Joint Institute for Nuclear Research, 29 June - 11 July

Spinodal Instabilities at the Deconfinement Phase Transition

Jørgen Randrup (Berkeley)

Possible (ρ , T) phase diagram of strongly interacting matter





Jørgen Randrup I

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Lecture I:	Phase coexistence (equilibrium)	WED 12:00-13:00
Lecture II:	Phase separation (non-equilibrium)	THU 16:30-17:30
	Discussion	18:00-19:00

Lecture III: Effects on collision dynamics (clumping) FRI 11:00-12:00



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Lecture I: Phase coexistence (equilibrium)



Thermodynamics: statistical mechanics of large *uniform* systems



Non-uniform systems: gradient effects, interface tension

Basic thermodynamics

$$\begin{array}{c} \begin{array}{c} 1\\ 1\\ \end{array} \\ \begin{array}{c} X_{1} = \{E_{1}, N_{1}, V_{1}, \ldots\} => S_{1}(X_{1}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} => S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} == S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} == S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} == S_{2}(X_{2}) \\ \end{array} \\ \begin{array}{c} X_{2} = \{E_{2}, N_{2}, V_{2}, \ldots\} \\ \end{array} \\$$

Thermodynamics with no conserved flavor

Statistical equilibrium in bulk matter

Control parameter(s) {X}:

Energy $E = V \varepsilon$ $\varepsilon = E/V$ Volume $V \rightarrow \infty$



Entropy function *S*{*X*}:

$$S(E,V) = V\sigma(\varepsilon)$$

Derivative(s) $\lambda_{\chi} = \partial_{\chi}S$: $\begin{cases} \beta = 1/T = \partial_{E}S(E, V) = \partial_{\varepsilon}\sigma(\varepsilon) & temperature \\ \pi = p/T = \partial_{V}S(E, V) = \sigma - \beta\varepsilon & pressure \end{cases}$

Thermodynamic (local) <u>stability</u>: $\delta^2 S_{tot} < 0$ => Entropy curvature $\partial_{\epsilon}^2 \sigma$ must be *negative*

Thermodynamic <u>coexistence</u>: => $T_1 = T_2$ & $p_1 = p_2$





Phase transformation with no conserved flavor



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Thermodynamics with one conserved flavor

Statistical equilibrium in bulk matter

Energy
$$E = V\epsilon$$
 $\epsilon = E/V$
Flavor $N = V\rho$ $\rho = N/V$



Entropy function *S*{*X*}:

Control parameter(s) {X}:

 $S(E, N, V) = V\sigma(\varepsilon, \rho)$

 $\beta = 1/T = \partial_E S(E, N, V) = \partial_\varepsilon \sigma(\varepsilon, \rho)$

Derivative(s) $\lambda_x = \partial_x S$:

$$\alpha = -\mu/T = \partial_N S(E, N, V) = \partial_\rho \sigma(\varepsilon, \rho)$$

$$\pi = p/T = \partial_V S(E, N, V) = \sigma - \beta \varepsilon - \alpha \rho$$

Thermodynamic <u>coexistence</u>: $\delta S_{tot} = 0 =>$ $T_1 = T_2 \& \mu_1 = \mu_2 \& p_1 = p_2$

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Thermodynamic (local) <u>stability</u>: $\delta^2 S_{tot} < 0 \Rightarrow$ Curvature matrix $\{\partial_{\chi}\partial_{\chi'} \sigma(\varepsilon,\rho)\}$ has only *negative* eigenvalues:

 $\begin{array}{ccc} \partial_{\varepsilon}^{2}\sigma & \partial_{\rho}\partial_{\varepsilon}\sigma \\ \partial_{\varepsilon}\partial_{\rho}\sigma & \partial_{\rho}^{2}\sigma \end{array}$

<u>Microcanonical scenario:</u> E and N are specified:



Example: Nuclear matter



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Nuclear phase diagram in different representations



Isentropic changes

Entropy density: $\sigma(\varepsilon,\rho)$

Energy density: ε Net baryon density: ρ Temperature: $T(\varepsilon,\rho) = 1/\sigma_{\varepsilon}$ Chemical potential: $\mu(\varepsilon,\rho) = -T\sigma_{\rho}$ Pressure: $p(\varepsilon,\rho) = T\sigma - \varepsilon + \mu\rho$ Enthalpy density: $h(\varepsilon,\rho) = p + \varepsilon$

Entropy per (net) baryon: $s(\varepsilon, \rho) = \sigma/\rho$

Changes: $(\delta \varepsilon, \delta \rho) \Rightarrow \delta s$:

 $\rho^{2}T\delta s = \rho^{2}T\delta(\sigma/\rho) = \rho T\delta\sigma - T\sigma\delta\rho = \rho\delta\varepsilon - \mu\rho\delta\rho - [h-\mu\rho]\delta\rho = \rho\delta\varepsilon - h\delta\rho$

$$\delta s = 0 \implies \rho \delta \varepsilon = h \delta \rho \iff \delta \varepsilon / \delta \rho = h / \rho$$

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Isentropic phase trajectories in different representations



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Example: Nambu – Jona-Lasino model

C. Sasaki, B. Friman, K. Redlich, Phys. Rev. Lett. 99, 232301 (2007): Density fluctuations in the presence of spinodal instabilities

$$\mathcal{L} = ar{\psi}(i \partial \!\!\!/ - m + \mu \gamma_0) \psi + G_S \Big[ig(ar{\psi} \psi ig)^2 + ig(ar{\psi} i ec{ au} \gamma_5 \psi ig)^2 \Big] \, .$$



Canonical description: T specified





Phase coexistence \Leftrightarrow common tangent:





Liquid-gas phase coexistence

uniform gas phase



uniform liquid phase



can *coexist* in mutual equilibrium

mixture



Thermodynamics of non-uniform matter



Thermodynamics of non-uniform matter: microcanonical

 $N = \int \tilde{
ho}(\boldsymbol{r}) d\boldsymbol{r}$ $E = \int \tilde{arepsilon}(\boldsymbol{r}) d\boldsymbol{r}$ $S = \int \tilde{\sigma}(\boldsymbol{r}) d\boldsymbol{r}$ Non-uniform flavor density $\tilde{\rho}(\boldsymbol{r})$ $\tilde{\varepsilon}(\boldsymbol{r})$ Non-uniform energy density Non-uniform entropy density $\ ilde{\sigma}[ilde{
ho}(m{r}), ilde{arepsilon}(m{r})](m{r})$ $\begin{cases} \tilde{T}(\boldsymbol{r}) = 1/\beta(\boldsymbol{r}) \\ \tilde{\mu}(\boldsymbol{r}) = -\tilde{\alpha}(\boldsymbol{r})\tilde{T}(\boldsymbol{r}) \end{cases}$ $\delta S = \int [ilde{eta}(m{r})\delta ilde{arepsilon}(m{r}) + ilde{lpha}(m{r})\delta ilde{
ho}(m{r})]dm{r}$ $\forall \, \delta \tilde{\varepsilon}(\boldsymbol{r}), \forall \, \delta \tilde{\rho}(\boldsymbol{r}): \ 0 \doteq \delta S - \beta_0 \delta E - \alpha_0 \delta N = \int [(\underbrace{\tilde{\beta}(\boldsymbol{r})}_{-\beta_0}) \delta \tilde{\varepsilon}(\boldsymbol{r}) + (\underbrace{\tilde{\alpha}(\boldsymbol{r})}_{-\alpha_0}) \delta \tilde{\rho}(\boldsymbol{r})] d\boldsymbol{r}$ $\forall \boldsymbol{r}: \quad \tilde{\beta}(\boldsymbol{r}) \doteq \beta_0 \quad \Rightarrow \quad \boldsymbol{\nabla} \tilde{\beta} \doteq \boldsymbol{0}$ Constant temperature: $\forall \boldsymbol{r}: \ \tilde{\alpha}(\boldsymbol{r}) \doteq \alpha_0 \ \Rightarrow \ \boldsymbol{\nabla} \tilde{\alpha} \doteq \boldsymbol{0}$ Constant chemical potential:

Constant pressure: $\begin{aligned}
\delta \pi &= -\varepsilon \delta \beta - \rho \delta \alpha & \pi \equiv p/T = \sigma - \beta \varepsilon - \alpha \rho \\
\mathbf{\nabla} \tilde{\pi} &= -\tilde{\varepsilon} \mathbf{\nabla} \tilde{\beta} - \tilde{\rho} \mathbf{\nabla} \tilde{\alpha} & \Rightarrow \tilde{p}(\mathbf{r}) = p_0
\end{aligned}$

J. Randrup, Phys. Rev. C 79, 054911 (2009)

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Thermodynamics of non-uniform matter: canonical

Constant temperature T Non-uniform flavor density $\tilde{\rho}(\mathbf{r})$ Non-uniform free-energy density $\tilde{f}_T[\tilde{\rho}(\mathbf{r})](\mathbf{r})$ $F_T = \int_{T} F_T = \int_{T}$

$$N = \int_{\Gamma} \tilde{\rho}(\boldsymbol{r}) d\boldsymbol{r}$$
$$F_T = \int_{\Gamma} \tilde{f}_T(\boldsymbol{r}) d\boldsymbol{r}$$

$$\delta F_T = \int \delta \tilde{f}_T(\mathbf{r}) d\mathbf{r} = \int \tilde{\mu}_T(\mathbf{r}) \delta \tilde{\rho}(\mathbf{r}) d\mathbf{r}$$
$$\forall \delta \tilde{\rho}(\mathbf{r}): \ 0 \doteq \delta F_T - \mu_0 \delta N = \int (\tilde{\mu}_T(\mathbf{r}) - \mu_0) \delta \tilde{\rho}(\mathbf{r}) d\mathbf{r}$$

Constant chemical potential: $\forall r: \tilde{\mu}_T(r) \doteq \mu_0 \Rightarrow \nabla \tilde{\mu}_T(r) \doteq 0$

Constant pressure:

$$\delta p = \rho \delta \mu$$

 $\nabla \tilde{p}_T(\mathbf{r}) = -\tilde{\rho}(\mathbf{r}) \nabla \tilde{\mu}_T(\mathbf{r}) \implies \tilde{p}_T(\mathbf{r}) = p_0$

J. Randrup, Phys. Rev. C 79, 054911 (2009)

H. Heiselberg et al., Phys. Rev. Lett. 70, 1355 (1993)

Finite range: gradient term

Free energy density for *uniform* matter at temperature *T*: $f_T(\rho)$

But we need to treat *non*-uniform systems: $\tilde{\rho}(\boldsymbol{r})$ Local-density approximation: $\tilde{f}_T[\tilde{\rho}(\boldsymbol{r})](\boldsymbol{r}) \approx f_T(\tilde{\rho}(\boldsymbol{r}))$

... implies:

$$F_{T}(\underline{ }) = F_{T}(\underline{ })$$

No good! => Finite range *must* be taken into account

Interface tension

$$\tilde{f}_T(x) = f_T(\tilde{\rho}(x)) + \frac{1}{2}C(\partial_x \tilde{\rho}(x))^2$$

The diffuse interface adds free energy (relative to a sharp interface); the excess in free energy (per unit area) is the *interface tension* γ_{τ} :

 $\left(\gamma_T^{12} = \int_{-\infty}^{+\infty} 2\Delta f_T(\tilde{\rho}(x)) dx = \int_{\rho_1}^{\rho_2} \left[2C\Delta f_T(\rho)\right]^{1/2} d\rho\right)$



$$dx = d\rho/\partial_x \rho$$

The interface profile $\rho(\mathbf{x})$ is **not** needed for γ_{τ} only the EoS for uniform matter, $f_{\tau}(\rho)$

The profile $\rho(x)$ is determined by: $C\partial_x^2 \tilde{\rho}(x) = \mu_T(\tilde{\rho}(x)) - \mu_0 = -\partial_\rho \Delta f_T(\tilde{\rho}(x))$

A recent example:

Marcus B. Pinto, V. Koch, and JR: Phys. Rev. C 86, 025203 (2012): Surface tension of quark matter in a geometric approach

Nambu -- Jona-Lasino model:

$$\mathcal{L}_{\text{NJL}} = \bar{\psi}(i\gamma_{\mu}\partial^{\mu} - m)\psi + G[(\bar{\psi}\psi)^{2} + (\bar{\psi}i\gamma_{5}\vec{\tau}\psi)^{2}]$$



$$\gamma_T = a \int_{\rho_1(T)}^{\rho_2(T)} [2\mathcal{E}_g \Delta f_T(\rho)]^{1/2} \frac{d\rho}{\rho_g}$$



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