



System for Continuous Chemical and Isotopic Purification of Hydrogen for the MuCap Experiment

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Why we need ultra pure hydrogen?

☐ Heavier elements

Negative muons preferentially transfer from μp atoms to heavier elements with high rates. Once a muon has transferred to a μN_z atom, nuclear muon capture proceeds more than 100 times faster than on a μp atom.

Thus, even tiny amounts of impurities in the hydrogen gas distort the observed μp lifetime spectrum.

Consequently, muon transfer and subsequent capture must be suppressed by keeping the gas contaminants below a level of **10 ppb**.

☐ Deuterium

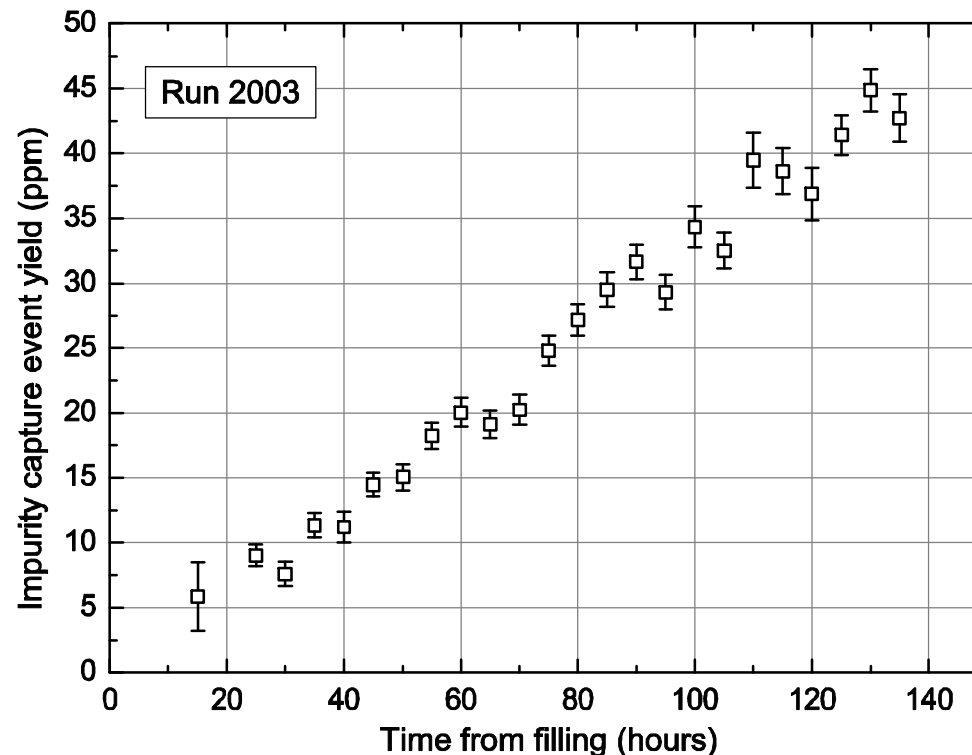
Isotopically pure hydrogen is required, since muons transfer to deuterium, where they pose a systematic problem for the MuCap experiment.

Due to a **Ramsauer-Townsend** minimum in the $\mu d + p$ scattering cross-section, μd atoms can diffuse over macroscopic distances and can either escape from the stopping volume in the TPC in a time-dependent way and can even reach the chamber materials, where muons are quickly captured.



Why we need continuous purification?

Impurity capture events measured **before purification system installation**, shows the impurity capture event yield increasing in the hours after filling the hydrogen vessel through the palladium filter (better than 1 ppb).





Reasons for Circulation system?

- constant flux is necessary for the permanent gas purification.
about 3 l/min
- price of pure hydrogen is essential.
about 1000 Eur/m³
- stable pressure in TPC is important.
10 bar +/- 0.1% (10 mbar)

Type of compressor? Normal or Cryo-?

10 bar absolute pressure

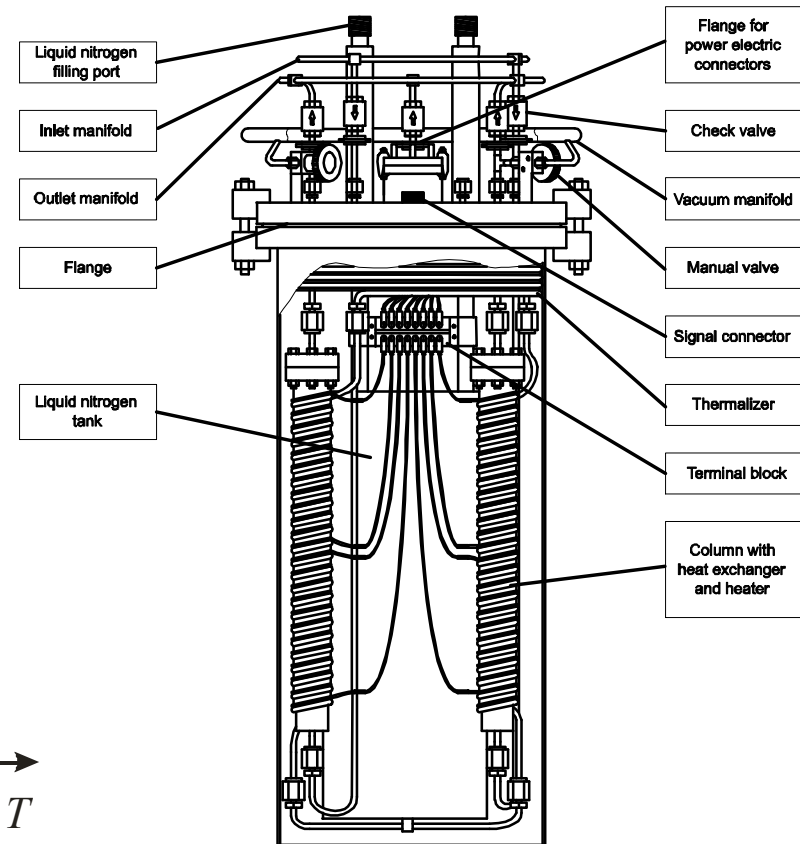
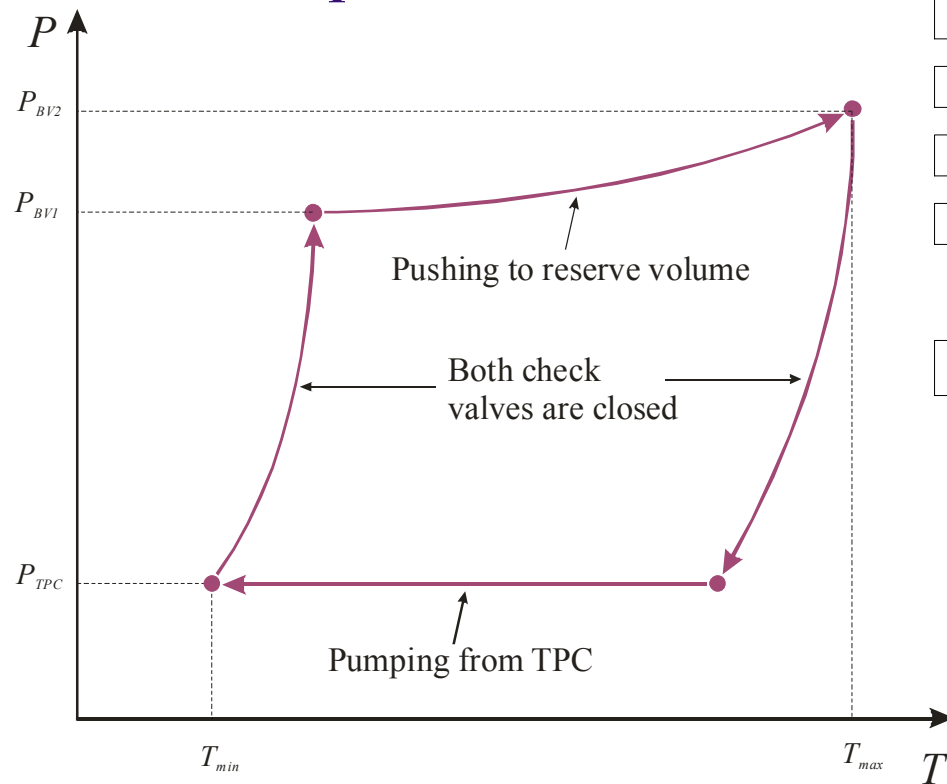
3 bar differential pressure

3-4 l/min flux

ultra pure!!!

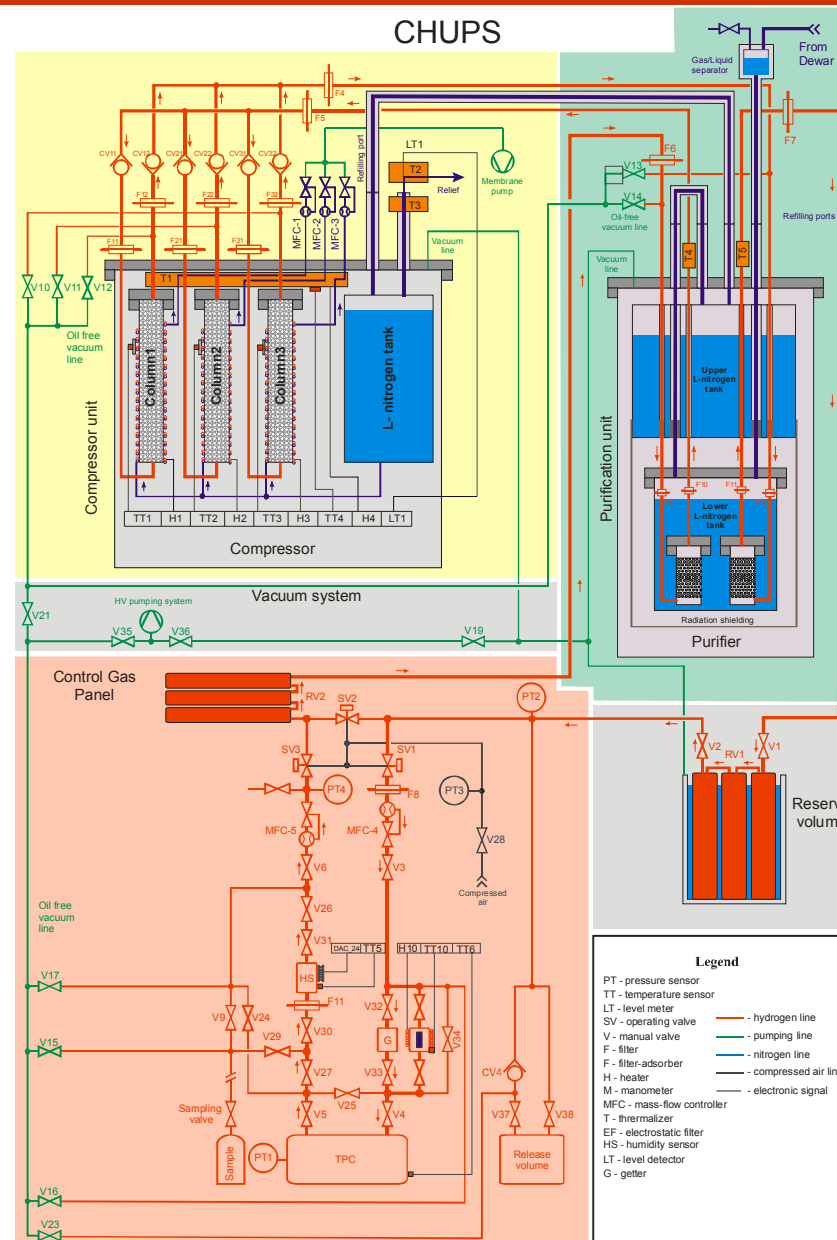


Simplified P-T diagram of a Compressor column



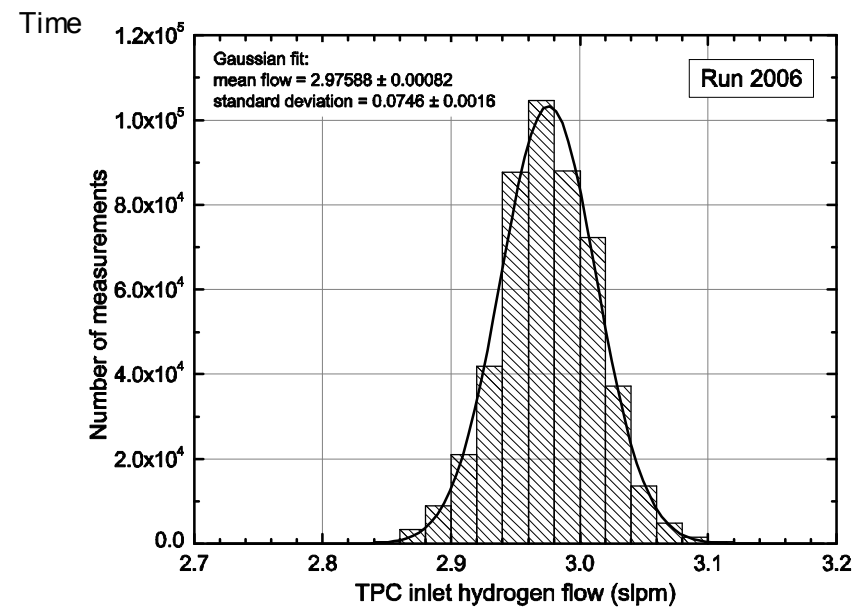
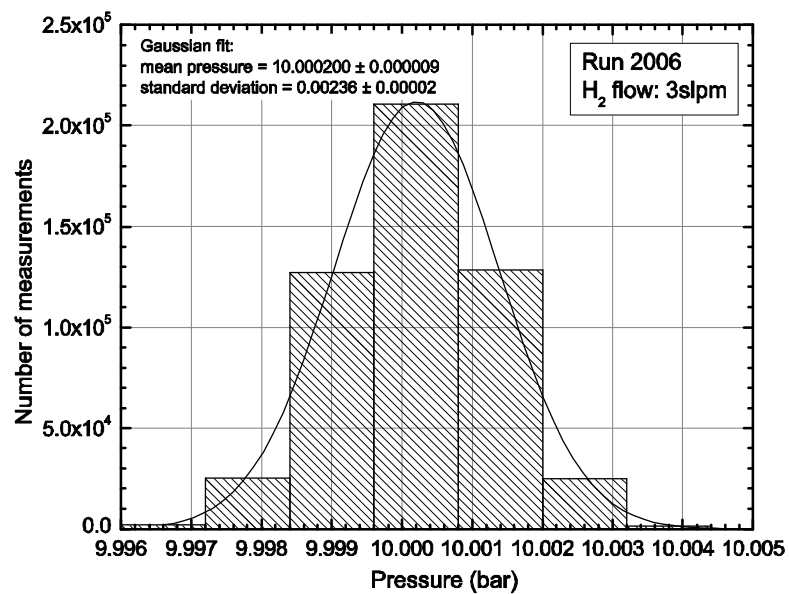
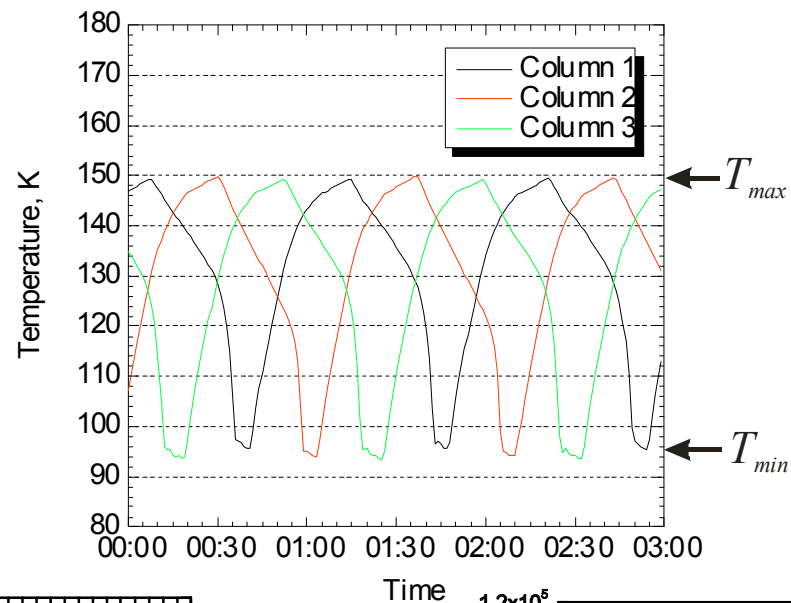


CHUPS scheme





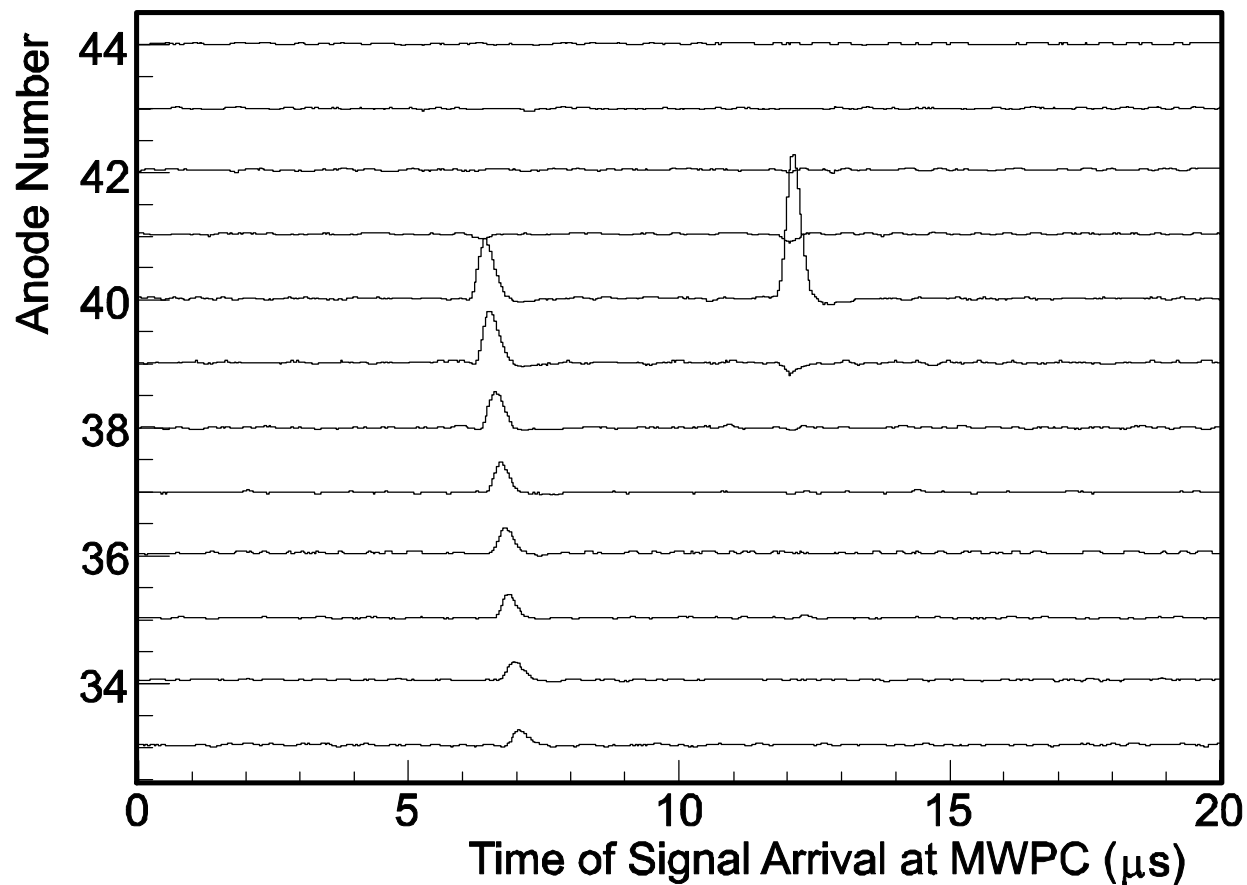
CHUPS, Hydrogen flux 3 l/min





Nitrogen purification

The capture process $\mu N_Z \rightarrow N_{Z-1} + \nu$ is identified by its distinct signature in the TPC

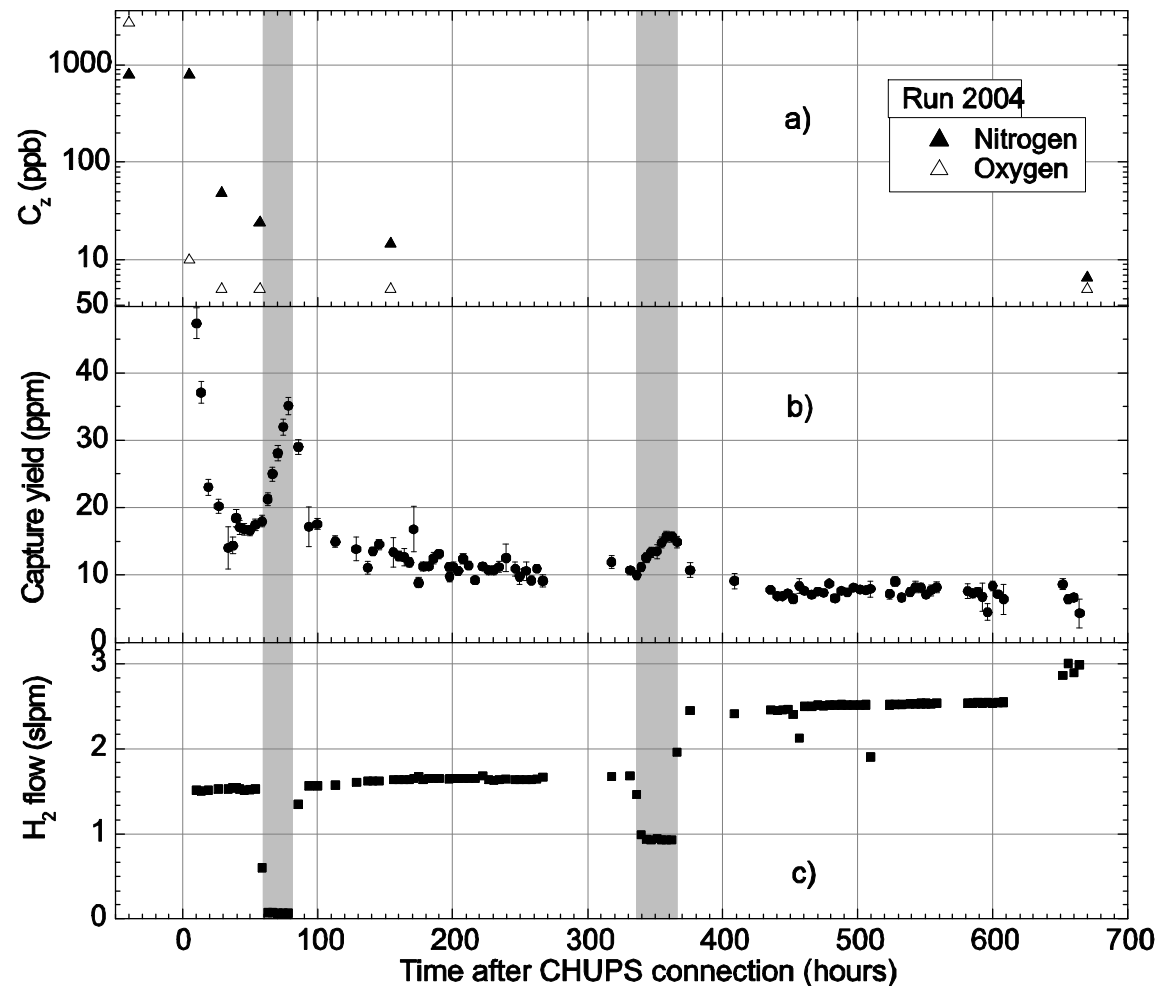


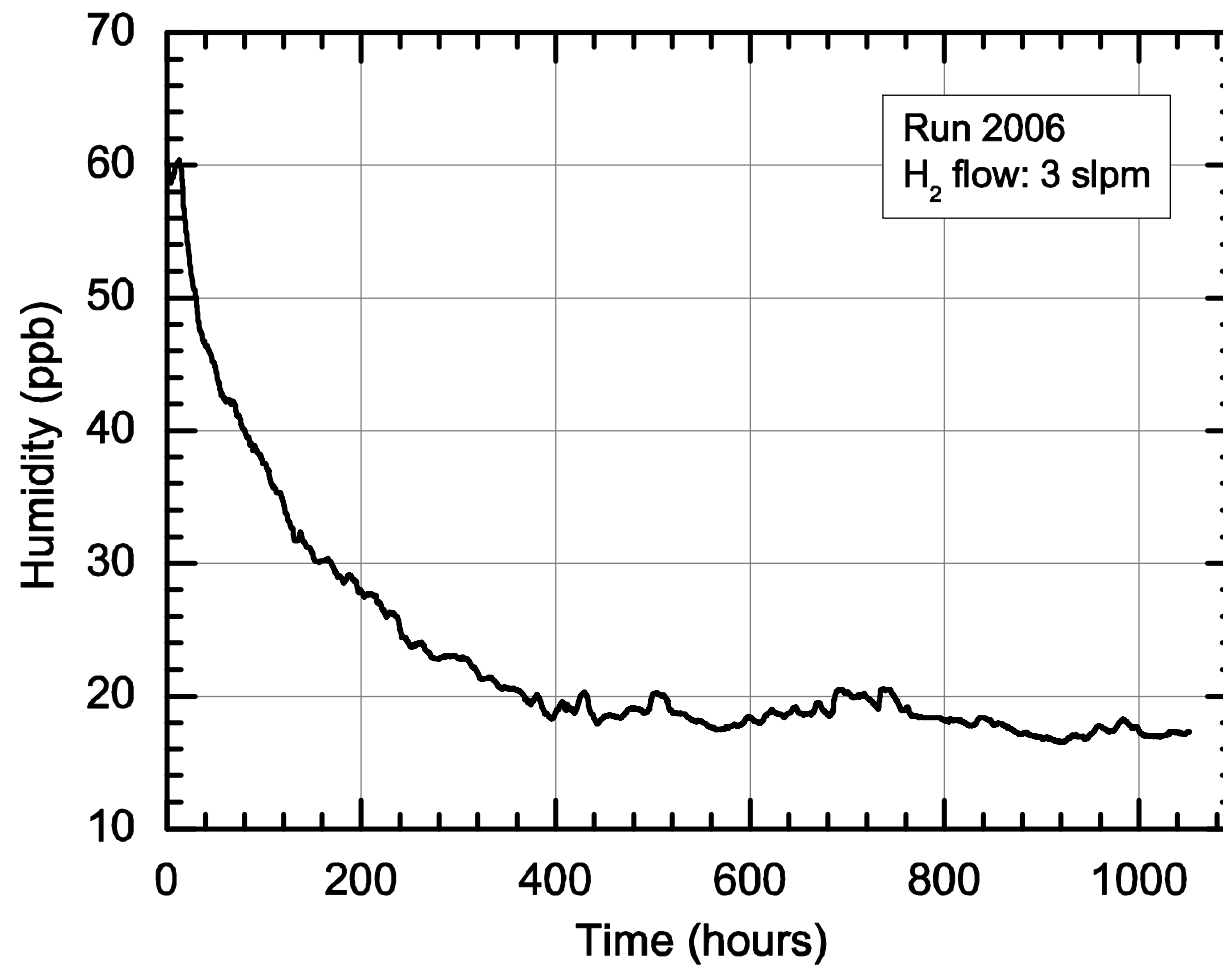


Nitrogen+Oxygen

Chromatography measurements also track the purification process (-a).
Oxygen traces dropped below the apparatus' sensitivity of **5 ppb** within two days after starting the circulation.

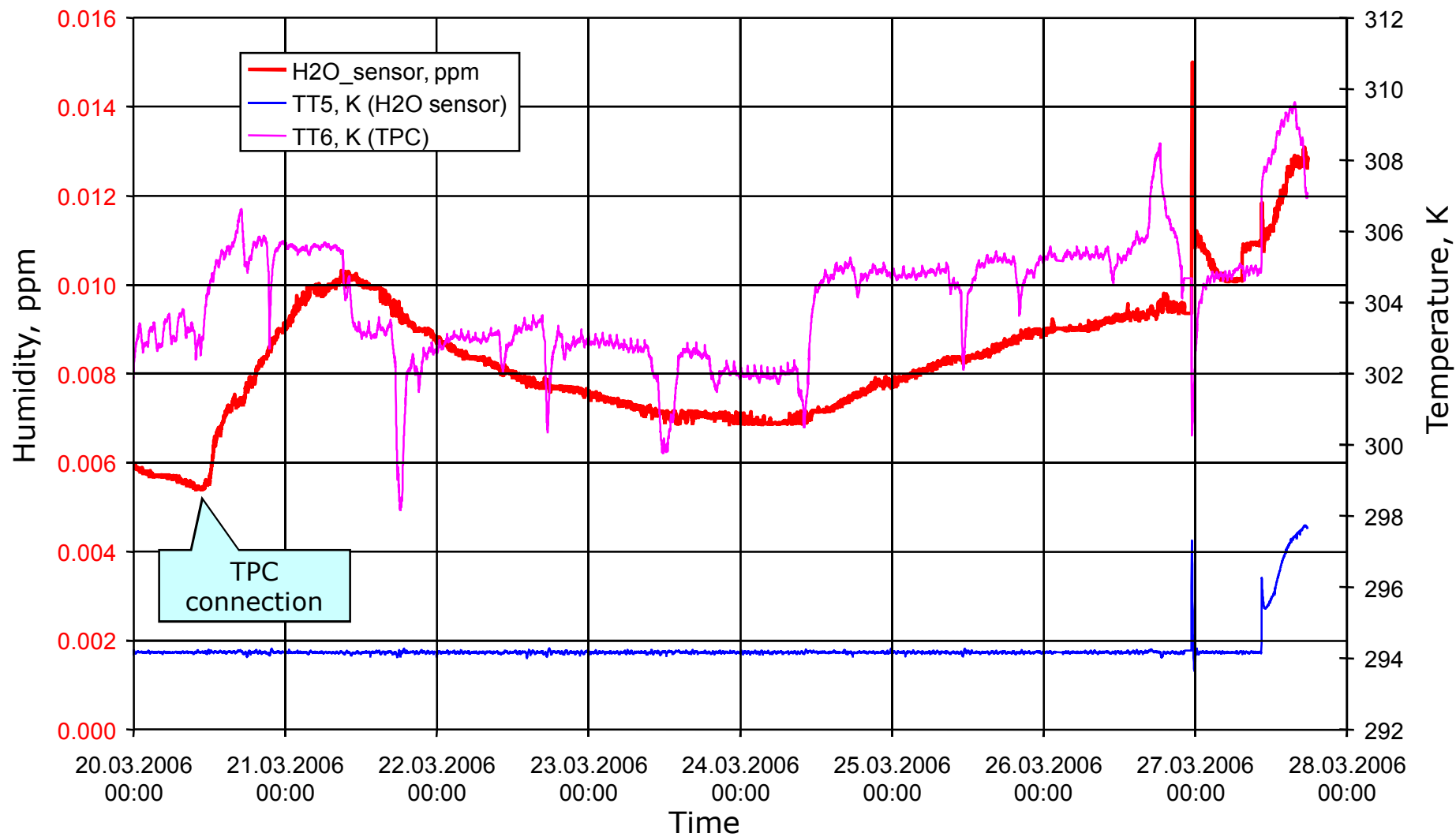
Nitrogen concentrations below the apparatus' sensitivity of **7 ppb**.







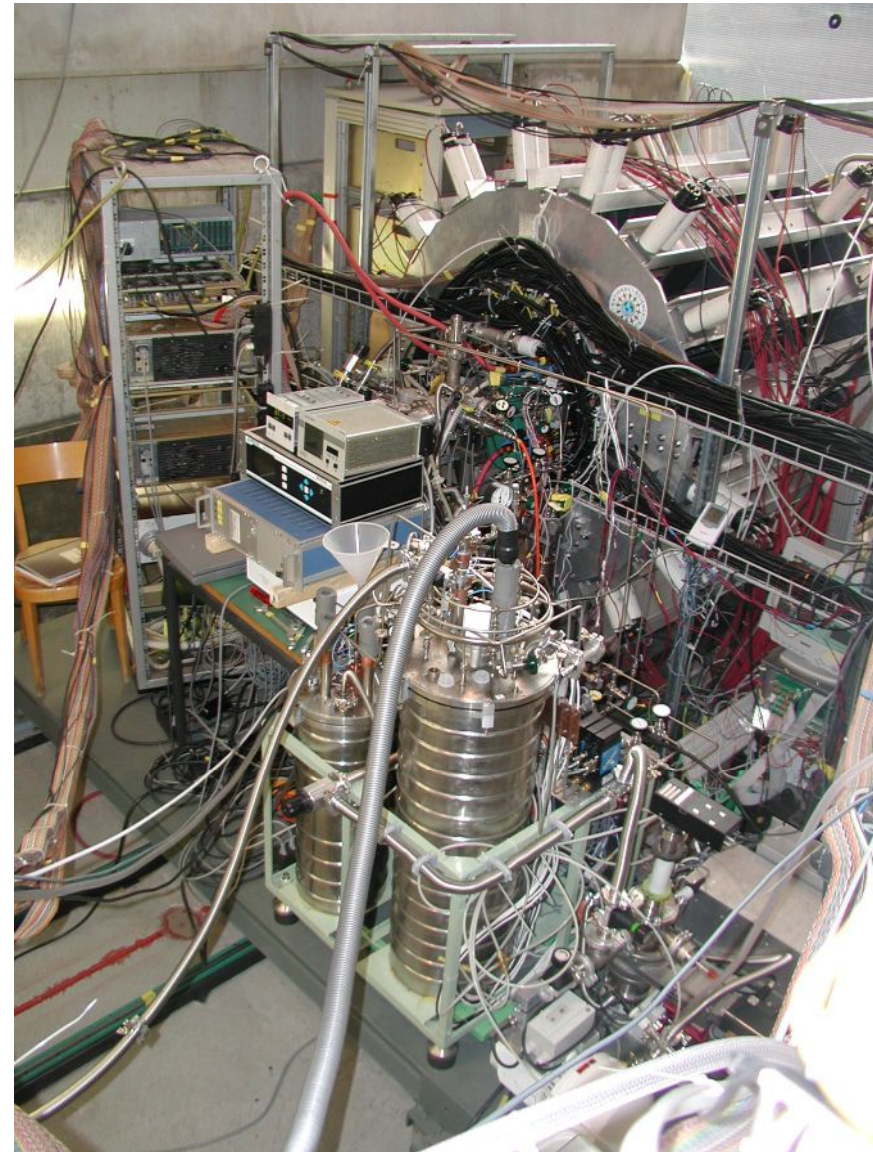
Moisture measurements in CHUPS with TPC





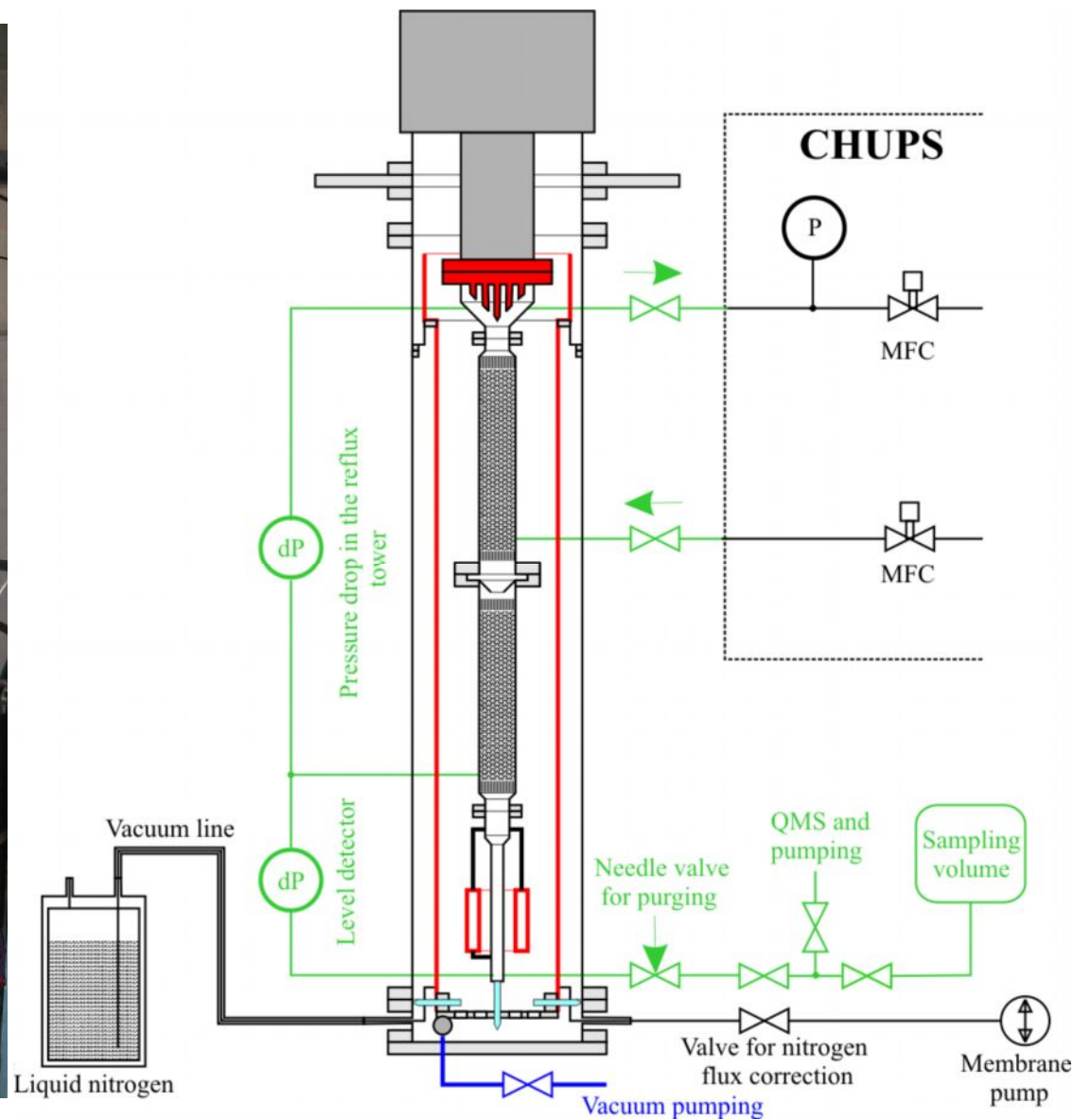
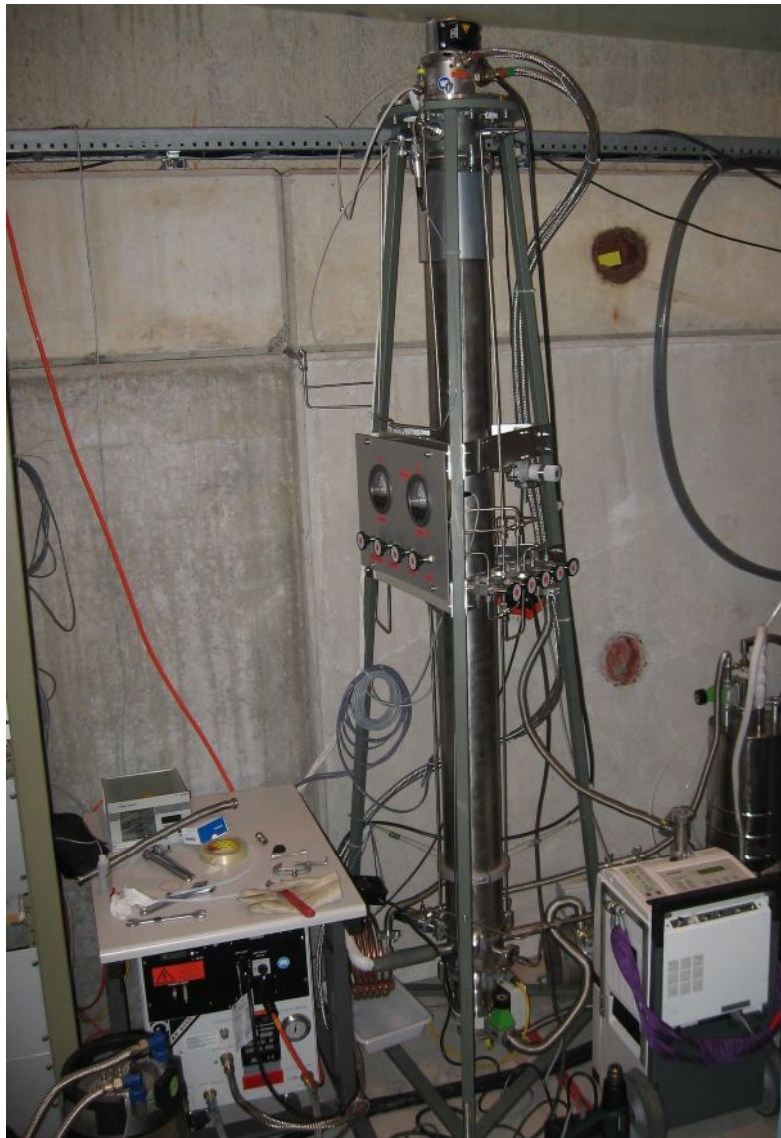
CHUPS results

N_2 less than 5-7 ppb
 O_2 less than 5 ppb
 H_2O about 20 ppb



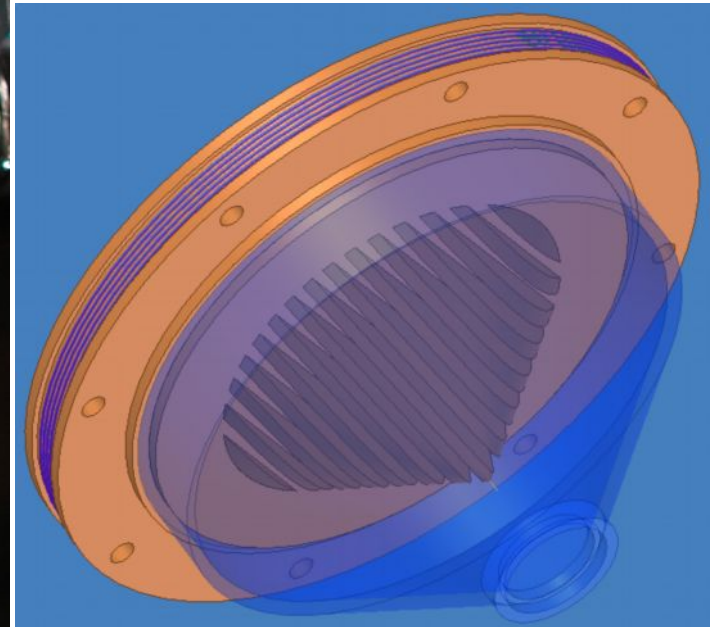
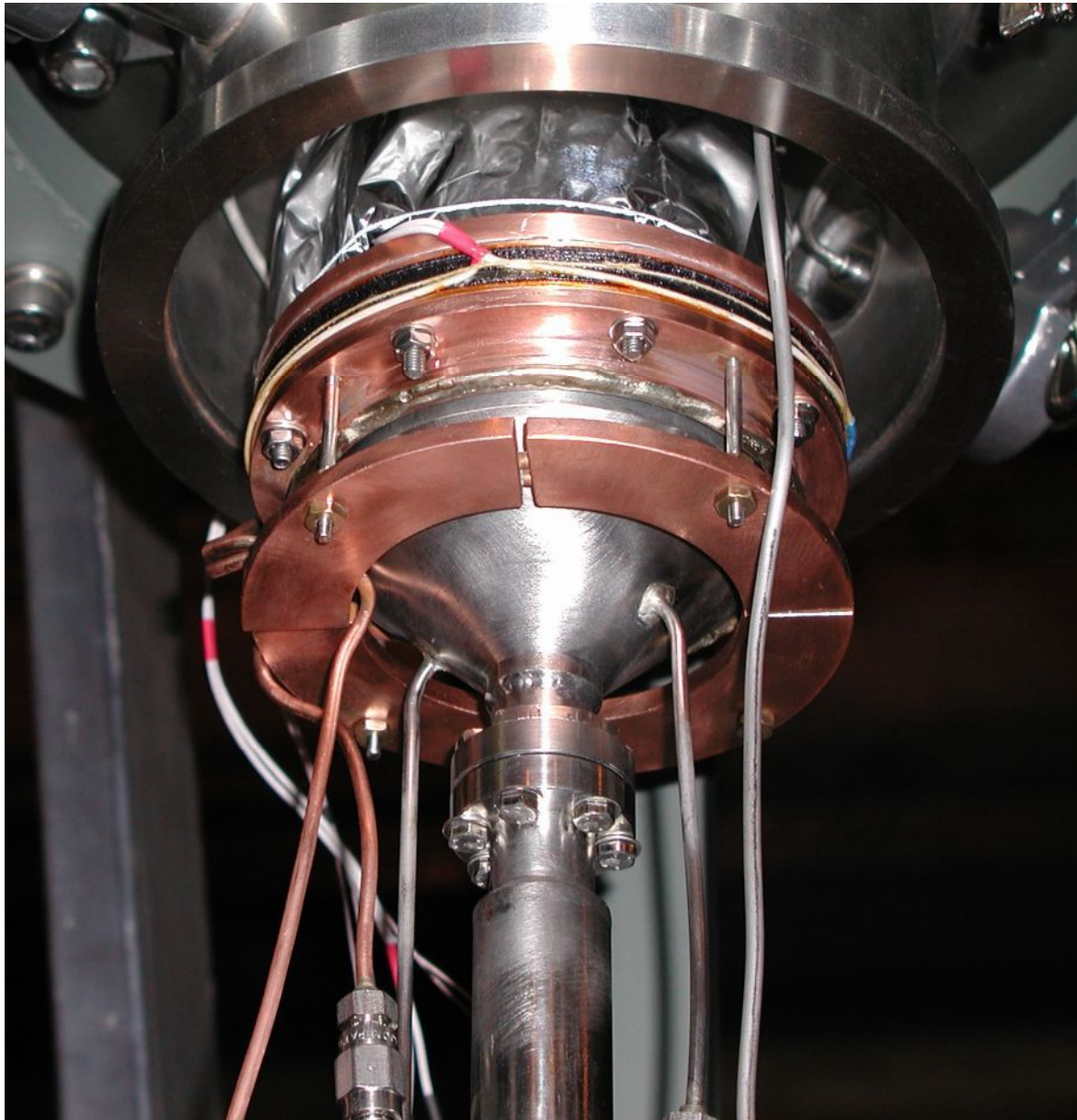


General layout of the cryogenic column



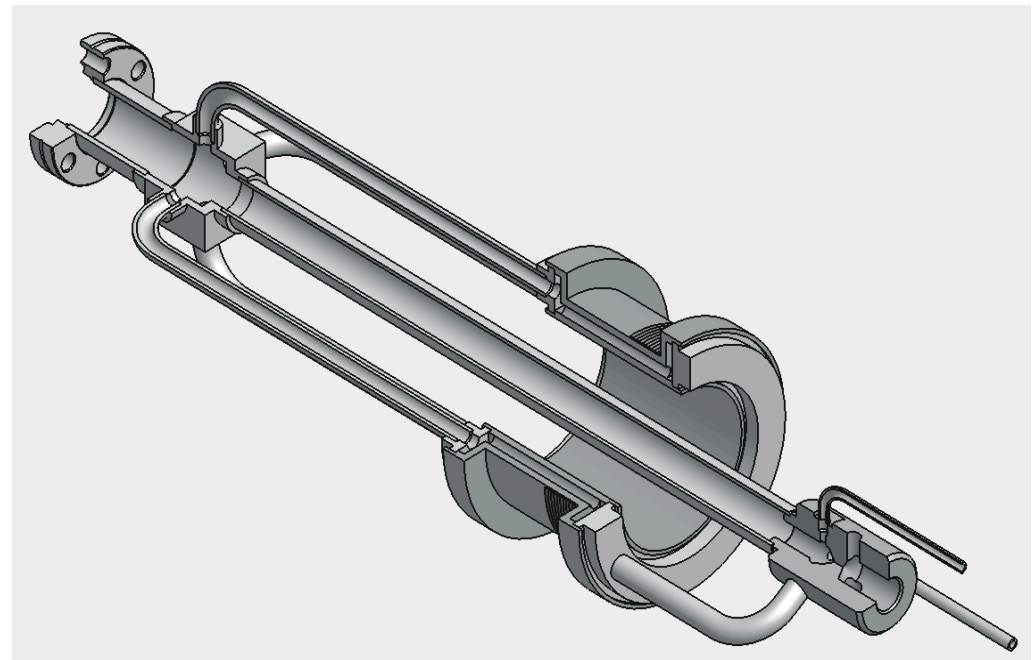
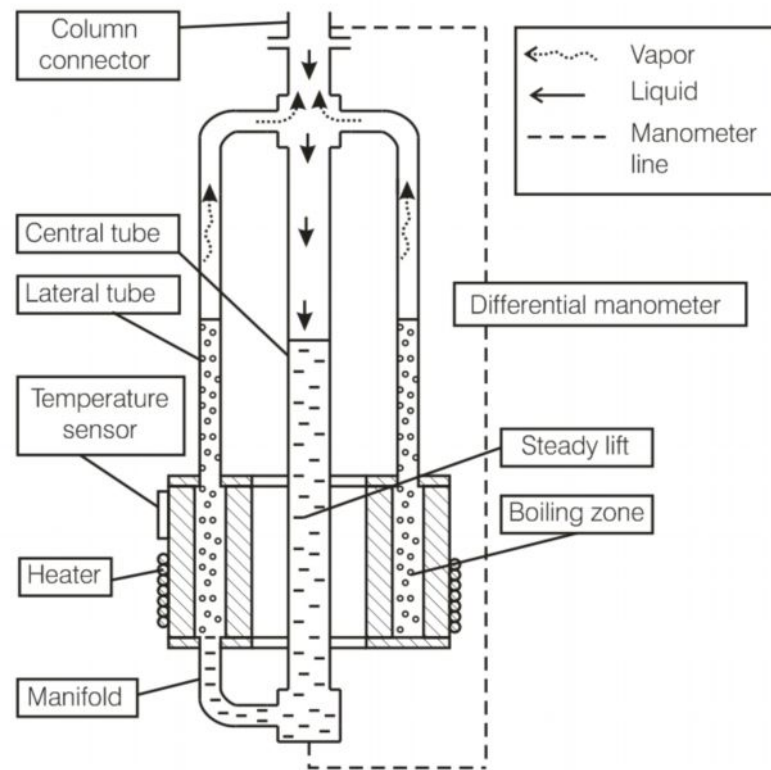


Condenser



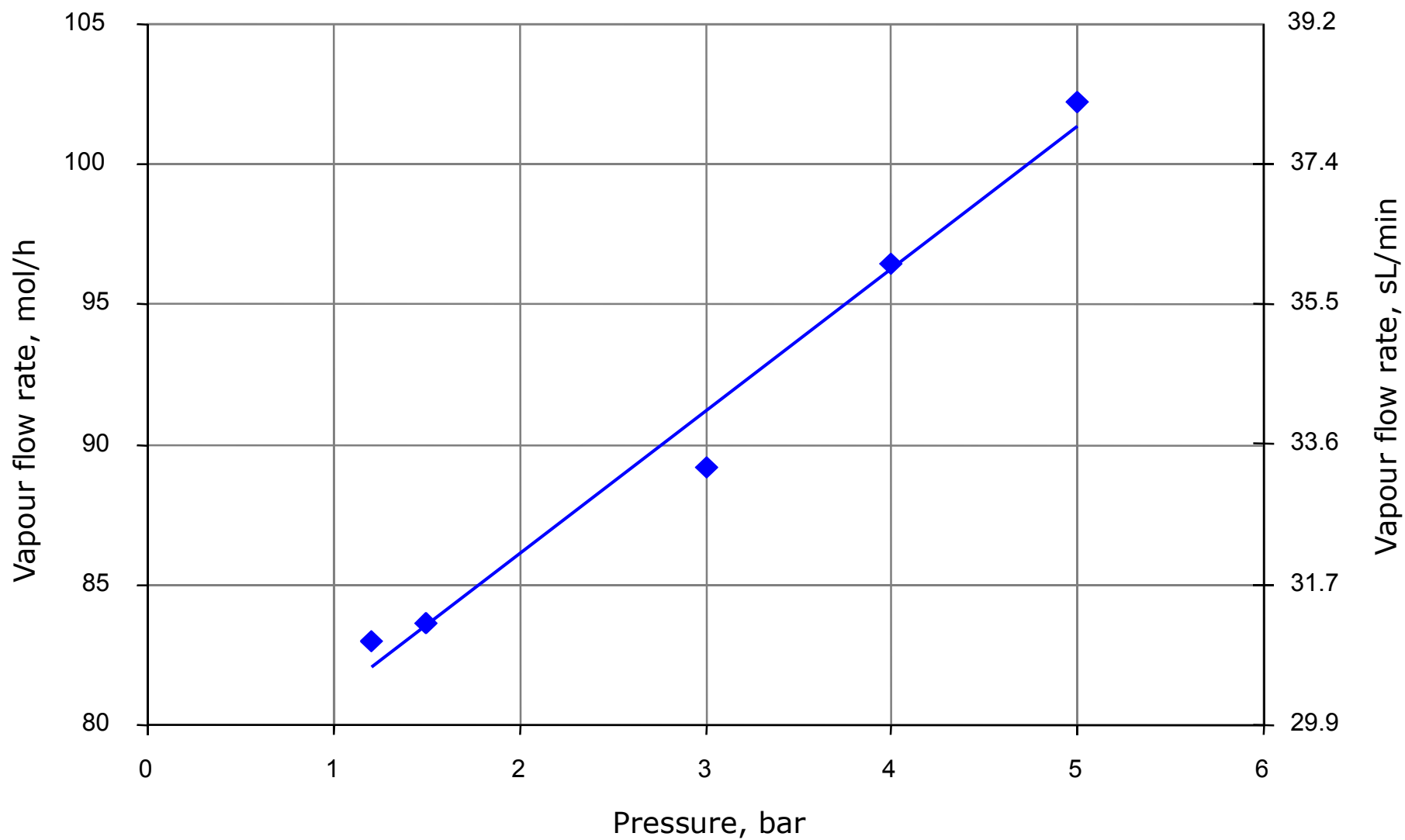


Reboiler



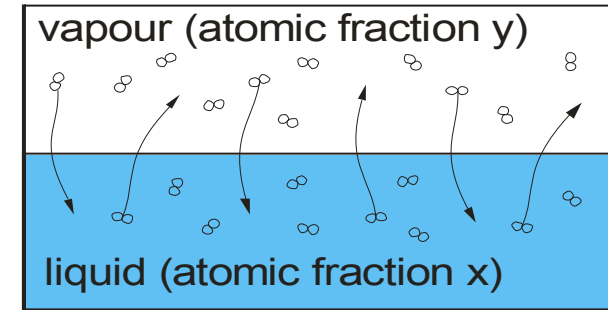
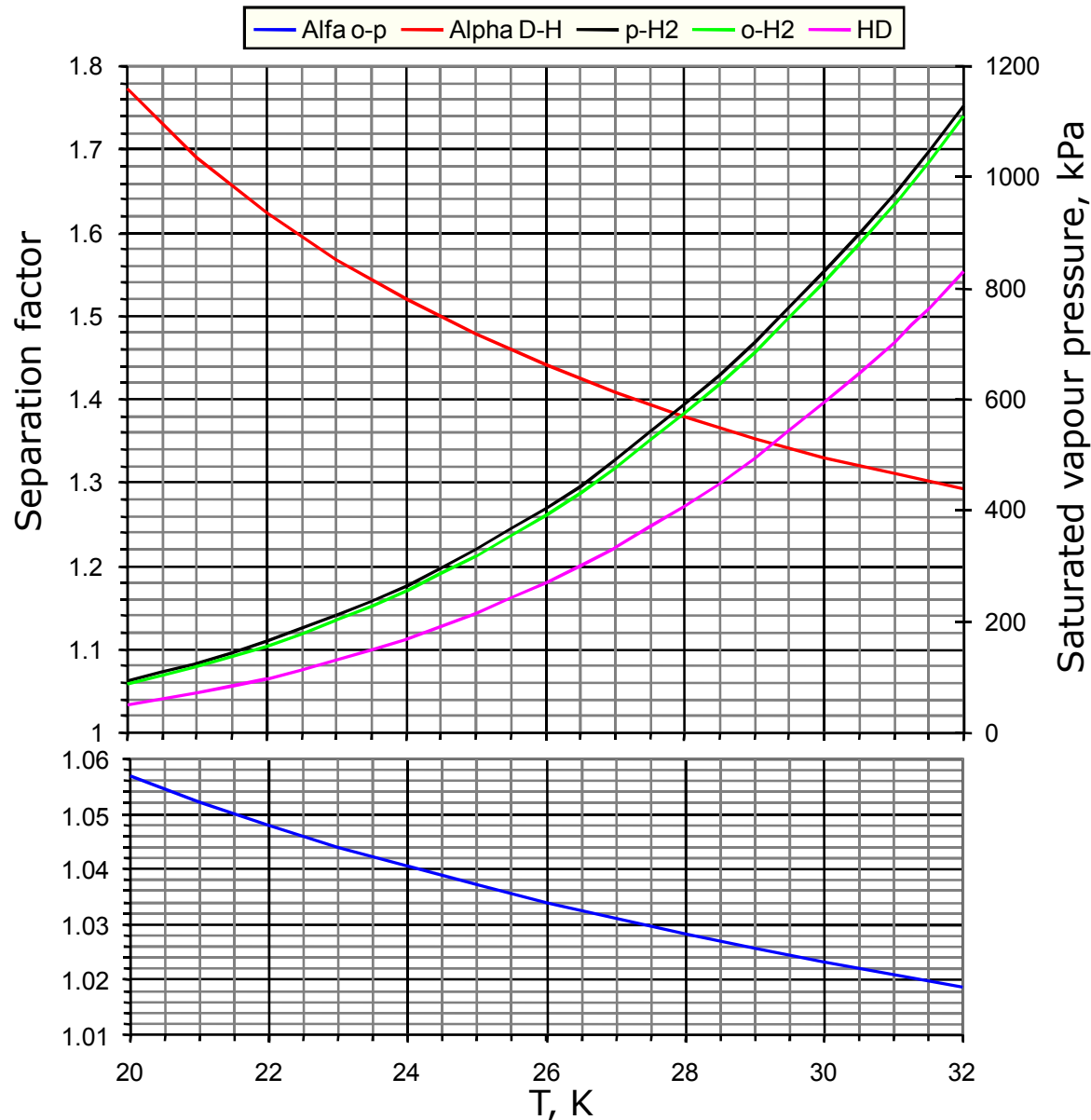


Vapour flow in the column at reboiler power $\sim 20\text{W}$





Saturated vapour pressures and separation factors



Temperature = const
 At dynamic equilibrium
 the content of low-boiling
 component is α -times higher
 in liquid.

Separation factors:

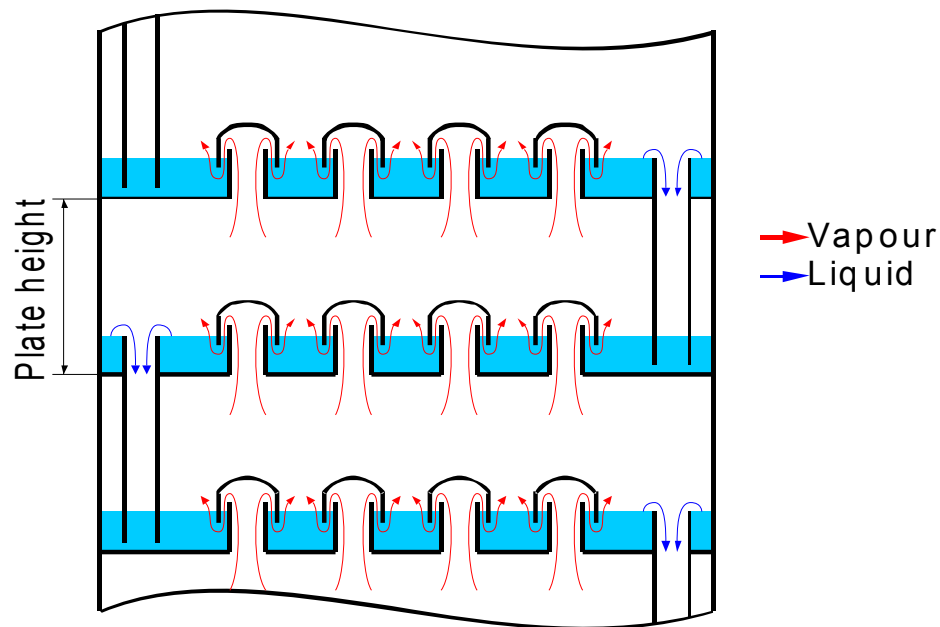
$$\alpha_{o-p} = \frac{P_{Sat.Para}}{P_{Sat.Ortho}}$$

$$\alpha_{D-H} = \frac{P_{Sat.H_2}}{P_{Sat.HD}}$$



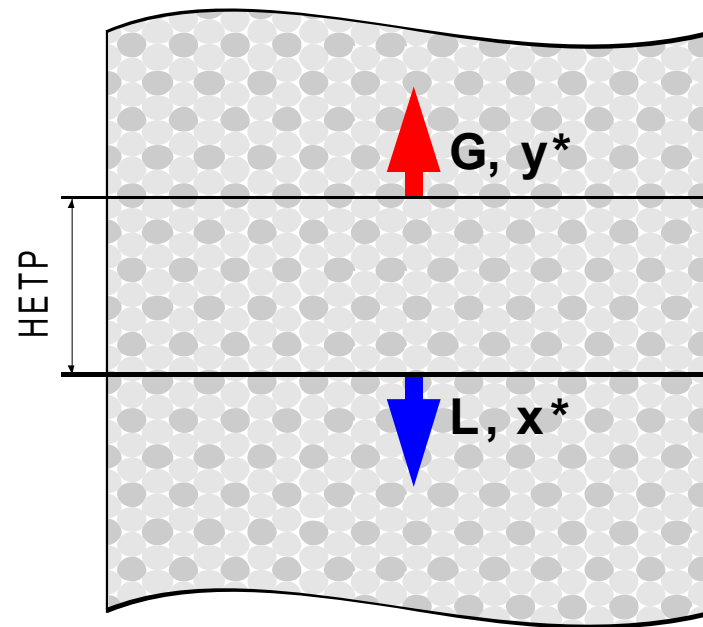
Height equivalent to a theoretical plate (HETP)

Bubble cap plate



Typical plate height ~15 cm
Efficiency = 70%

Non-regular packing



Height Equivalent to a Theoretical Plate (HETP)

Theoretical plate – a part of the column, where composition of outgoing liquid and vapour flows are in equilibrium.
(Abstract equivalent of the bubble cap plate with efficiency = 100%)

Atomic fractions:
x - in liquid
y - in vapour

Separation factor:

$$\alpha = \frac{x^* / (1 - x^*)}{y^* / (1 - y^*)}$$
 * indicates equilibrium composition



Packing

Characteristics:

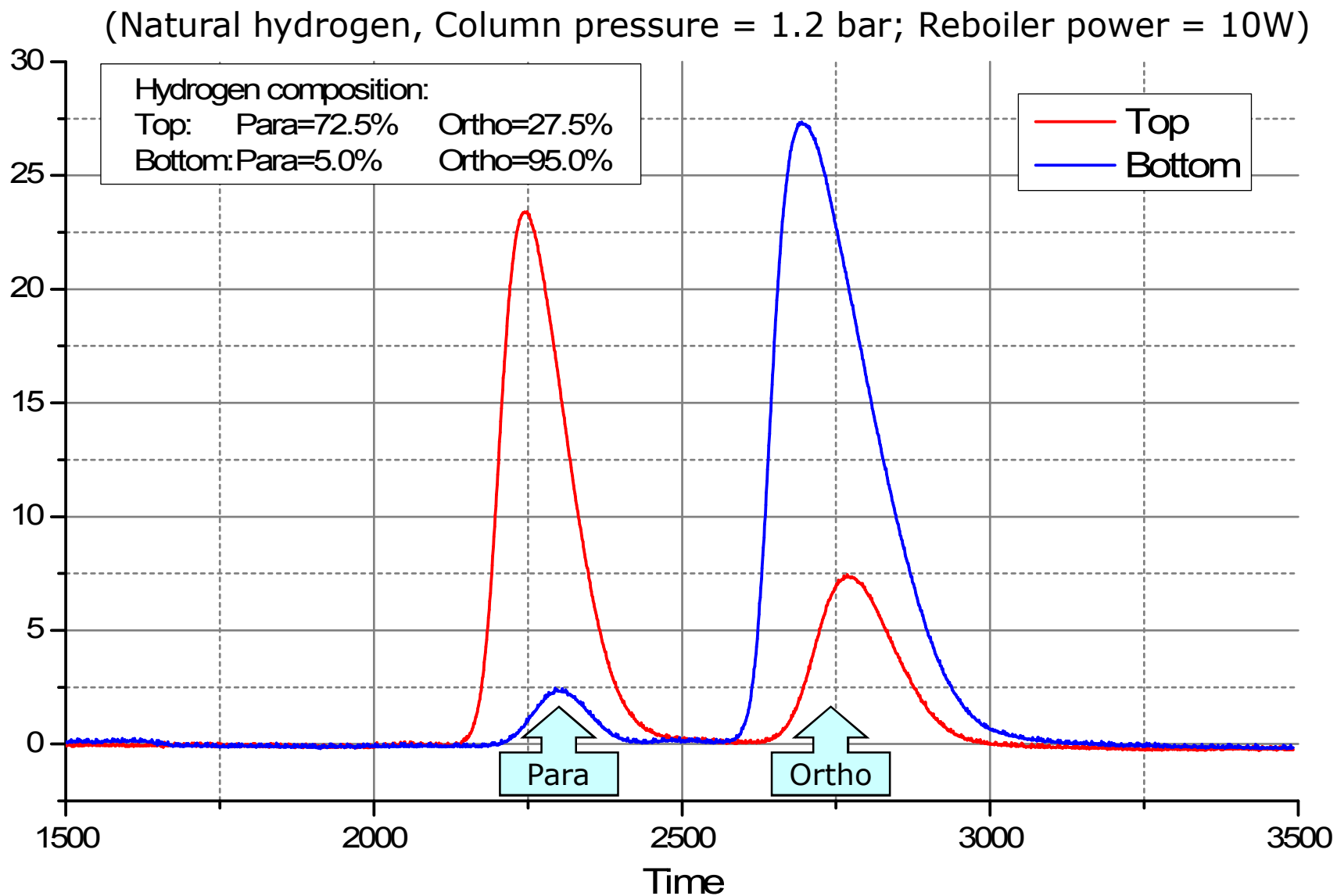
- Type: spiral prismatic
- Size: 2x2x0.2 mm
- Free volume fraction: 0.82
- Specific surface: 3490 m²/m³
- Packed density: 1430 kg/m³
- Material: stainless steel

Total volume of packing
in the column = 560 ml
Total packing surface
in the column = 1.95 m²



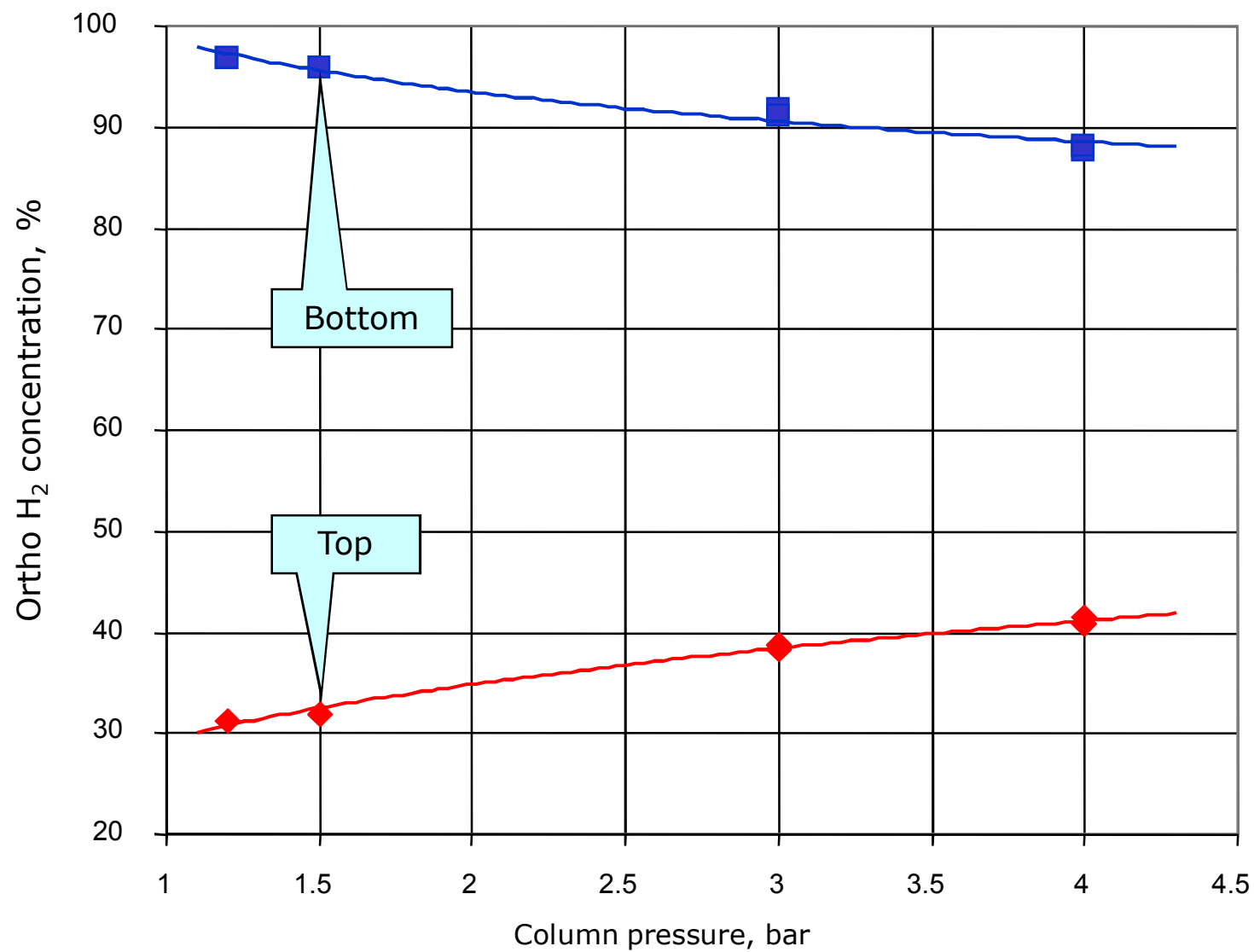


Ortho-Para Hydrogen Chromatogram



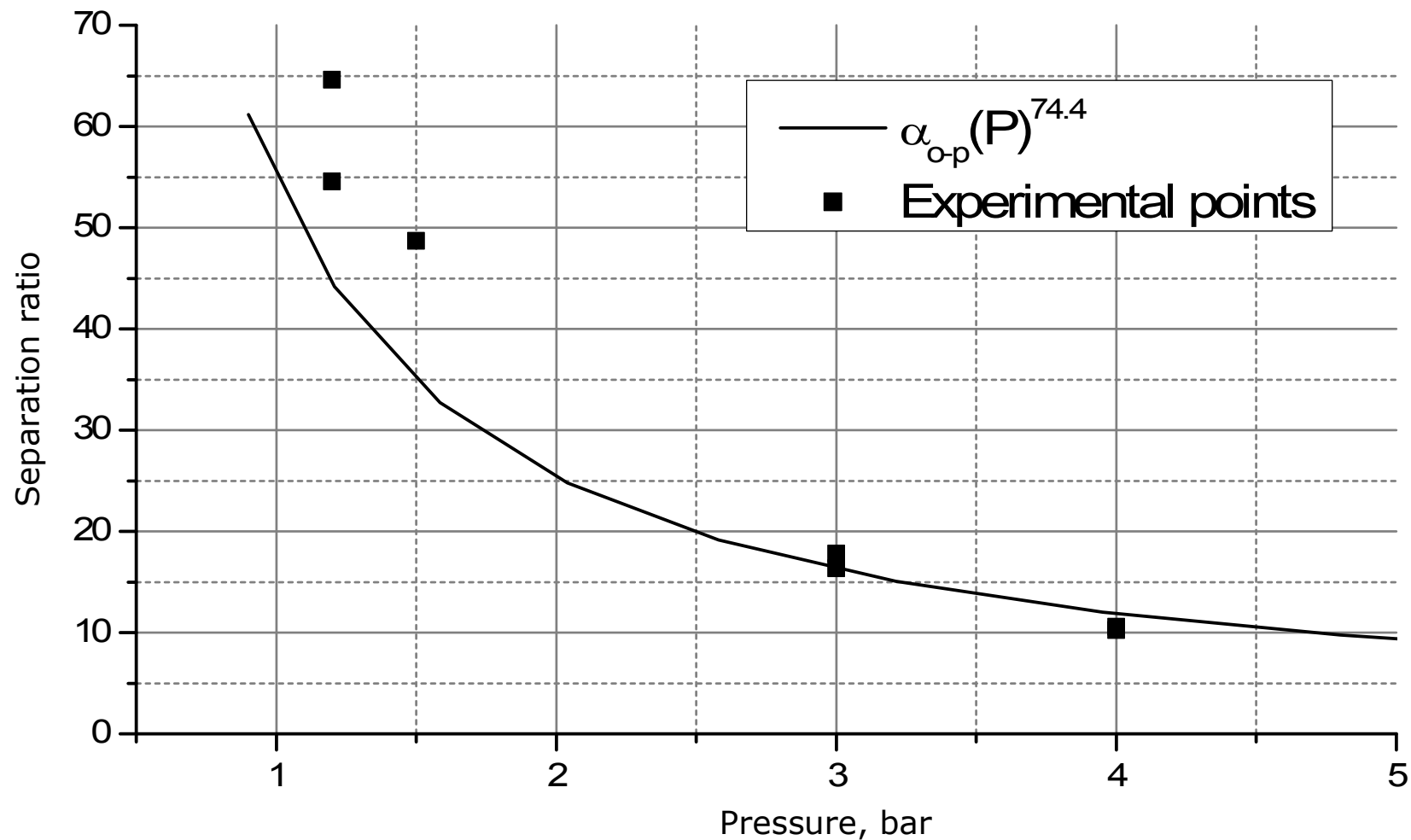


Ortho hydrogen concentration





Separation ratio (SR)

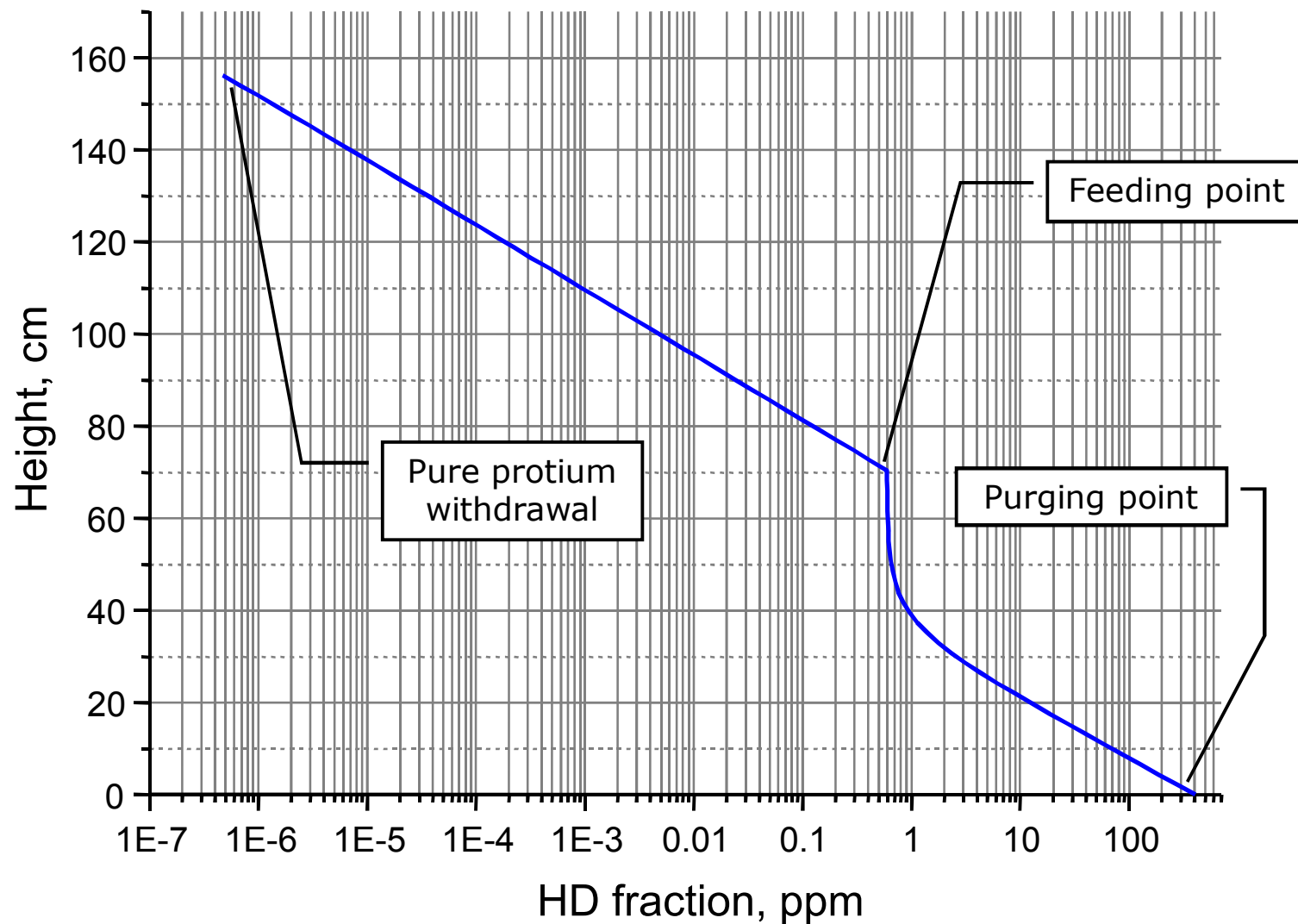


$SR = \alpha^N$ - Fenske equation for total reflux mode
(N - the number of theoretical plates)
HETP=Packing height / N

$$SR = \frac{X_{Bottom} / (1 - X_{Bottom})}{X_{Top} / (1 - X_{Top})}$$



Expected HD concentration profile



Feed flow rate = 1 L/min; Pressure = 1.5 bar; Purging flow rate = 0.015 L/min; HETP = 2.2 cm
Initial HD concentration = 6 ppm (deuterium atomic fraction = 3 ppm)



Deuterium purification results

The final measurements of the probes were performed on the new 200 kV Tandem accelerator built for isotope analysis in Zurich.

A special ion source was constructed giving extremely low backgrounds of hydrogen ions from walls, etc. The walls are continuously sputtered to keep the background low allowing measurements during 2 hours.

The existence of the zero samples from the DRU system turned out to be crucial, since the accelerator gives a different background if the ion source is not fed with hydrogen gas.

First zero sample measurement gave zero deuterium concentration at **70 ppb** sensitivity.

Thus, it contains less than **70 ppb** of deuterium.



Conclusions

- ❑ Separation column performance:
 - ❑ Height Equivalent to a Theoretical Plate (HETP) value of **2.2 cm** corresponds to the best medium power cryogenic column results. HETP value is almost constant in wide range of vapour flow rate.
 - ❑ Output deuterium concentration better than **70 ppb** does not depend on the initial concentration (natural hydrogen can be used!).
- ❑ It is possible to produce pure ortho- or para- hydrogen (99%) in the continuous mode.